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Enhancement of benzothiazoles as Pteridine Reductase-1 (PTR1) inhibitors for the treatment of Trypanosomatidic infections.

Pasquale Linciano^{1*}, Cecilia Pozzi², Lucia dello Iacono², Flavio di Pisa², Giacomo Landi², Alessio Bonucci², Sheraz Gul³, Maria Kuzikov³, Bernhard Ellinger³, Gesa Witt³, Nuno Santarem⁴; Catarina Baptista⁴, Caio Franco⁵, Carolina B. Moraes⁵, Wolfgang Müller⁶, Ulrike Wittig⁶, Rosaria Luciani¹, Antony Sesenna¹, Antonio Quotadamo¹, Stefania Ferrari¹, Ina Pöhner⁷, Anabela Cordeiro-da-Silva⁴, Stefano Mangani^{2*}; Luca Costantino^{1*}, Maria Paola Costi^{1*}.

¹University of Modena and Reggio Emilia, Via Campi 103, 41125 Modena, Italy

²University of Siena, Dipartimento di Biotecnologie, Chimica e Farmacia, Via Aldo Moro 2, 53100 Siena, Italy
³Fraunhofer Institute for Molecular Biology and Applied Ecology Screening Port, 22525 Hamburg, Germany
⁴Institute for Molecular and Cell Biology, 4150-180 Porto, Portugal and Instituto de Investigação e Inovação em Saúde, Universidade do Porto and Institute for Molecular and Cell Biology, 4150-180 Porto, Portugal
⁵Laboratório Nacional de Biociências (LNBio), Centro Nacional de Pesquisaem Energia e Materiais (CNPEM), 13083-100 Campinas – SP, Brazil; Present Address: Department of Microbiology, Institute of Biomedical Sciences, University of Sao Paulo, 05508-900 Sao Paulo – SP, Brazil.
⁶Scientific Databases and Visualization Group, Heidelberg Institute for Theoretical Studies (HITS), D-69118

Heidelberg, Germany.

⁷Molecular and Cellular Modeling Group, Heidelberg Institute for Theoretical Studies (HITS), D-69118 Heidelberg, Germany.

Abstract

2-amino-benzo[d]thiazole was identified as new scaffold for the development of improved PTR1 inhibitors and anti-Trypanosomatidic agents. Molecular docking and crystallography guided the design and synthesis of 42 new benzothiazoles. The compounds were assessed for *T. brucei* and *L*.

major PTR1 inhibition and *in-vitro* activity against *T. brucei* and amastigote *L. infantum*. We identified several 2-amino-benzo[d]thiazoles with improved enzymatic activity (*Tb*PTR1 IC₅₀ = 0.35 μ M; *Lm*PTR1 IC₅₀ = 1.9 μ M) and low μ M anti-parasitic activity against *T. brucei*. The ten most active compounds against *Tb*PTR1, were able to potentiate the antiparasitic activity of methotrexate when evaluated in combination against *T. brucei*, with a Potentiating Index between 1.2 and 2.7. The compound library was profiled for early ADME-Toxicity and 2-amino-*N*-benzylbenzo[d]thiazole-6-carboxamide (**4c**) was finally identified as a novel potent, safe and selective anti-trypanocydal agent (EC₅₀ = 7.0 μ M). Formulation of **4c** with hydroxypropyl- β -cyclodextrin yielded good oral bioavailability, encouraging progression to *in-vivo* studies.

 Kinetoplastidae are a group of flagellated protozoan parasites transmitted by insect vectors and responsible for severe human pathologies. Human African Trypanosomiasis (HAT; also known as African sleeping sickness), is caused by *Trypanosoma brucei* (*T. brucei*), the Chagas disease is caused by *Trypanosoma cruzi* (*T. cruzi*) and three different manifestations of leishmaniasis (visceral, cutaneous and mucocutaneous) are caused by infection with different species of *Leishmania*.¹ In particular, HAT can arise in an acute or chronic form. Cardiac and kidney diseases, among other serious complications, can lead to death if untreated. The second stage arises when the parasite migrates across the blood-brain barrier, resulting in neurological symptoms that ultimately can lead to coma and death.²

Although treatment options for these infections exist, their use is limited by several factors, including toxicity, poor efficacy, difficult administration and cost. Furthermore, increasing levels of drug resistance emphasize the need for new, improved and affordable drugs.^{3,4} Recently, after thirty years of research in this field, fexinidazole, a 5-nitroimidazole, has been finally approved as the first oral drug for the treatment of HAT.^{5,6}

In order to accelerate the discovery of new treatments for Trypanosomatidic infections, one approach already used for the treatment of bacterial infections and some parasitic diseases (e.g. malaria) entails the use of drugs targeting folate enzymes (e.g. thymidylate synthase, TS or dihydrofolate reductase, DHFR).^{7,8} As a result, the reduction of the cellular pool of deoxythymidine monophosphate, which is essential for DNA replication, causes cell death.⁸ Trypanosomatids are auxotrophic for folates and pterins; therefore, inhibiting the enzymes involved in the folate salvage pathways should provide effective treatment.⁹ However, the use of DHFR inhibitors (e.g. methotrexate (MTX) or pyrimethamine) for the treatment of these parasitic diseases is unsuccessful, pteridine reductase 1 (PTR1) activity.¹⁰ PTR1 is partly due to а short-chain dehydrogenase/reductase (SDR) that can reduce both conjugated and unconjugated pterins. Indeed,

expression of PTR1 is promoted when parasitic DHFR is inhibited, ensuring survival of the parasite.^{11–13} The entanglement between parasitic PTR1 and DHFR activity is corroborated by the limited anti-parasitic efficacy of full PTR1 inhibitors developed to date,^{14–19} and by the necessity to co-administer a DHFR inhibitor to observe a trypanocidal activity.^{15,17,20} Recently, *T. brucei* PTR1 (*Tb*PTR1) inhibitors based on structural elaboration of 2-amino-1,3,4-thiadiazoles were proposed.¹⁷ The new compounds showed only micromolar activity against *Tb*PTR1 and potentiated the activity of MTX when used in combination²⁴.

A double aromatic/heteroaromatic ring system was deemed necessary to implement the design of the thiadiazole core. A two-ring system is expected to provide improved PTR1 inhibitors with increased affinity for the protein, thus competing more strongly with the substrate pterin moiety. Since 2010, a number of X-ray crystal structures have been made solved in which two aromatic/heteroaromatic ring systems were bound to PTR1 as protein inhibitors mapping the active site.^{17,21–24} From our preliminary studies on compound repositioning, riluzole and other 2-amino-benzothiazole derivatives exhibited low micromolar affinity against PTR1.¹⁵ The aim of the present work was to enlarge the benzothiazole library through a structure-based approach, to characterize the biological activity against PTR1 and Trypanosomatidic parasites and evaluate the early ADME-Toxicity profile to select at least one compound for progression to pharmacokinetic evaluation in animal assays.

X-ray crystal structures with 2-amino-benzothiazole that could reveal the precise core active-site interactions were not available. Therefore, we first obtained the X-ray crystal structure of (6-(methylsulfonyl)benzo[d]thiazol-2-amine) with *Tb*PTR1 and, based on this structure, we designed and synthesized a library of 42 compounds. The selection and prioritization of compounds for *in-vivo* evaluation was performed following the criteria considered for the Target Inhibitor Profile (TIP) ^{25,26} resulting in one compound selected for pharmacokinetic studies.

RESULTS AND DISCUSSION

TbPTR1- NADP(H)-3a crystal structure

The chemical starting point of the present study relied on the benzothiazole **1b** and **3a** that originated from previous virtual screening studies.¹⁵As a result of their moderate inhibitor activity against *T. brucei* (IC₅₀ = 34.2 μ M and 15.1 μ M for **3a** and **1b**, respectively) and high ligand efficiency (LE = 0.32 for both **3a** and **1b**), the two fragments, together with the 2-aminobenzothiazole core scaffold, were considered to be important sub-structures at the outset of this work.



Figure 1. Chemical structures and enzymatic inhibitor activities against *T. brucei* and *Leishmania major* (*L. major*) PTR1 of the benzothiazoles fragments **1b** and **3a**.

Previous docking studies, performed on **1b** and **3a** against *L. major* PTR1 (*Lm*PTR1) showed binding at the active site with the benzothiazole core sandwiched between the cofactor nicotinamide and phenylalanine, forming an array of hydrogen bonds with catalytic residues and the cofactor phosphate and ribose.¹⁵ To provide crucial information that could assist the design of improved inhibitors, crystallographic studies were attempted. **1b** and **3a** were used for crystallization trials with *Tb*PTR1 and the structure of the ternary complex *Tb*PTR1–NADPH–**3a** was successfully obtained.

The structure of *Tb*PTR1 in complex with the cofactor NADP(H) and compound **3a** (IC₅₀ of 34.2 μ M, PDB ID 6GCL) was determined to 1.95 Å resolution (Figure 2). The crystal asymmetric unit includes the whole functional *Tb*PTR1 tetramer whose structure is highly conserved with those previously described in the literature^{17,24,27} (subunits were completely traced apart from two surface-exposed loops, including residues 104-112 and 143-15, that are usually poorly visible in *Tb*PTR1 structures). The 2-amino-benzothiazole core of **3a** occupies the catalytic cavity, as previously predicted by molecular modelling,¹⁵ in which it exploits its extended aromatic system to generate the peculiar π -sandwich interaction within Phe97 and the cofactor nicotinamide. A network of H-bonds entailed by the amine moiety, donating two hydrogen bonds to the cofactor β -phosphate, stabilizes the scaffold. In addition, hydroxyl of Ser95 and benzothiazole nitrogen both receives an H-bond from the hydroxyl of the cofactor ribose. The sulfonic group of **3a** is located in the active site pocket lined by the side chains of Cys168, Val206, Leu209, Pro210, Met213, and Trp221.



Figure 2. Active site view of *Tb*PTR1 (white cartoon, active site residues in sticks with white carbons) in complex with the cofactor NADP(H) (in sticks, black carbons) and **3a** (in sticks, purple carbons). The inhibitor is surrounded by the omit map (green wire) contoured at the 2.5 σ level. H-bonds are represented as red dashed lines. Atom color code: nitrogen (blue), oxygen (red) and sulfur (yellow). (PDB ID: 6GCL)

Benzothiazoles design

In accordance with the binding mode of the 2-amino-benzothiazole fragment observed in the X-ray structure of *Tb*PTR1-NADP(H)-**3a** and the architecture of the parasitic *Tb*PTR1, new PTR1 inhibitors based on the benzothiazole scaffold were designed and synthesized. The catalytic pocket of *Tb*PTR1 is located in the deepest part of the active site and it is delimited by Phe97, Ser95, Arg14, Asp161, Leu209, Pro210, Tyr174 and the nicotinamide moiety of NADPH. The entrance of the pocket is a hydrophobic tunnel (hydrophobic pocket 1, Figure 3) formed by Val206, Trp221 and Met213 where the *p*-amino benzoic acid (*p*ABA) ring of a folic acid substrate is accommodated (see PDB: 3BMC²⁸). Next to hydrophobic pocket 1, a second hydrophobic pocket is present (Figure 3), delimited by Cys168, Val206, Phe97 and closed on the top by Phe171.



Figure 3. *Tb*PTR1 active site. The catalytic, primary and secondary hydrophobic pockets are highlighted. The surface of the protein is colored based on the hydrophobicity of the amino acid residues (from red: hydrophobicity to white: hydrophilicity). The relevant residues delimiting the pocket and NADP(H) are displayed as sticks. (PDB ID: 6GCL)

An internal virtual library of around 500 synthetically accessible benzothiazole derivatives was rationally designed in a structure-based approach. The crystal structure of **3a** was taken as a template for the design; the 2-amino-benzothiazole core scaffold anchors the molecule within the catalytic pocket of the *Tb*PTR1 active site and during the design it was retained unaltered and decorated as reported in Figure 4. The compounds were designed based on the following considerations. Aliphatic or aromatic side chains were appended in position 6 of the benzothiazole ring to probe the hydrophobic pocket 1 or the secondary hydrophobic cleft (Figure 4A). The side-chain on C-6 of the aromatic substituent was linked to the main core scaffold by means of four different linkers, varying in flexibility, polarity, lipophilicity and length (e.g. ether, sulfide, sulfone and amide) as shown in Figure 4B. Moreover, more sterically hindered amidic derivatives were also explored to simultaneously fill both hydrophobic pockets (Figure 4C). Finally, hydroxyl groups were introduced at carbon 6 and carbon 7 on the benzothiazole core scaffold to explore the H-bond interactions with the residues of the active site (Figure 4D). The diverse substituents were properly combined to generate a library that was screened by molecular docking with Autodock 4.2 on the 3JQ7 crystal structure (see Supporting Information).^{28,29}



Figure 4. Main structural components of the 500 compounds virtual library based on the 2amino-benzothiazoles core.

The docking model was first validated by re-docking the DX2 inhibitor²¹ into the parent crystal structure with the resulting RMSD being within 0.83Å. Thereafter, the virtual benzothiazole based library was docked and docking poses were filtered first according to the predicted scores ($\Delta G_{binding}$ ranging from -6 to -9 kcal/mol). The ligand-protein complexes were then visually inspected and compounds able to reproduce the binding pattern of the 2-amino-benzothiazole core scaffold observed in the *Tb*PTR1– NADP(H)–3a crystal structure were selected for synthesis (series 1-4, Table 1-4). Figure 5 shows the most reliable predicted binding mode of the designed benzothiazoles addressed by the synthesis.



Figure 5. Predicted binding mode by docking calculation against *Tb*PTR1 of some representative designed benzothiazoles. A) Predicted pose of **2e** (green carbon) reproducing the binding mode of 2-amino-benzothiazole in the catalytic pocket and showing the placement of the aromatic side chain in the hydrophobic pocket 1. B) Superimposition of the ternary complex NADP⁺–*Tb*PTR1–Folate (PDB ID: 3BMC, folic acid in green carbons) and the predicted binding mode of **4g** (in orange carbons). C) Superposition of the predicted binding mode of four homologous benzothiazoles differing only for the linker (**1c**, **2c**, **3c** and **4c** respectively in magenta, cerulean, green and yellow carbons). D) Predicted binding mode of hydroxylated benzothiazole **5b** (in magenta carbons). Model atoms except for carbons are color coded with protein carbons (white), oxygen (red), nitrogen (blue), sulfur (yellow) and phosphorous (orange). The protein backbone is shown as white cartoon.

42 molecules with the best docking scores and poses were synthesized (Table 1-5). The four homologous series, in accordance with the linker type, were divided in etheric (**1a, c-g,** Table 1), sulfide (**2a-h,** Table 2), sulfone (**3a, c-g,** Table 3) and amide (**4c-h,** Table 4) benzothiazoles. Derivatives **1h**, **3b** and **4a-b** could not be synthesized. Assuming no more than one violation to the rule of 5, (molecular weight, number of H-bond donor and acceptor, number of rotatable bonds, *log* $P_{o/w}$ and polar surface area), all the designed compounds were in accordance with Lipinski's rule (Table SI-1). The compounds passed the check for pan-assay interference compounds (PAINS) evaluated with the *in silico* tool FAFdrugs4.^{30,31}

Chemistry

Etheric benzothiazoles **1a**, **c-g** were prepared as reported in **Scheme 1**. 6-hydroxy-2aminobenzothiazole (**6**) was synthesized first through condensation between benzoquinone and thiourea in refluxing ethanol in presence of concentrated hydrochloric acid as catalyst. **6** was further reacted in standard S_N2 condition with iodomethane (to prepare **1a**) or with the appropriate benzylbromide to obtain **1c-f**. **1g** was synthesized by hydrolyzing **1f** with 1N NaOH in THF:MeOH 3:1 at room temperature for 24 hours. The carboxylic acid thus obtained was further reacted with methyl isonipecotate via standard coupling condition with EDC, HOBt in DMF at room temperature for 24 hours to give **1g**.



Scheme 1. Reagents and conditions: (a) HCl conc. (0.1 eq.), EtOH, refl., 6 h. (b) K_2CO_3 (2.5 eq.), DMF, r.t., 12 h. (c) 1N NaOH aq. (1.5 eq.), THF:MeOH 3:1, r.t., 24 h. (d) methyl isonipecotate (1



Synthesis of 6-(sulfone)-2-aminobenzothiazoles **3a**, **c-g** was performed as reported in **Scheme 2**, passing by the synthesis of the relative sulfides **2a-h**. 4-nitrothiophenol was reacted first with the appropriate benzyl bromide and K₂CO₃ as base in DMF to afford 4-nitro-*S*-benzylbenzenes **7a-f** that were subsequently reduced to the respective anilines **8a-f** with zinc and ammonium chloride in refluxing methanol. Anilines **8a-f** were then cyclized to benzothiazoles **2a-f** with potassium thiocyanate in glacial acetic acid using elemental bromine as oxidizing agent. For the synthesis of sulfides **2g-h**, **2f** was hydrolyzed first to carboxylic acid and consequentially reacted in standard coupling condition with methyl-isonipecotate (to give **2g**) or benzylamine (to give **2h**). Sulfides **2a-g** thus obtained were further oxidized to the respective sulfones **3a**, **c-g** with *m*-CPBA in DCM at room temperature for 24 hours. It was not possible to synthesize sulfones **3b** and **3h** due to decomposition of the precursors **2b** and **2h** in presence of oxidizing agents.



Scheme 2. Reagents and conditions: (a) K₂CO₃ (2.5 eq.), DMF, r.t., 24 h. (b) Zn (2 eq.), NH₄Cl

(2.5 eq.), MeOH, rifl. 2 h. (c) Br_2 (2 eq.), KSCN (4 eq.), glacial acetic acid, T < 10 °C, 1-6 h. (d) 1N NaOH aq. (1.5 eq.), THF:MeOH 3:1, r.t., 24 h. (e) methyl isonipecotate (1 eq.) or benzylamine (1 eq.), EDC (1 eq.), HOBt (1 eq.), DMF, 0°C to r.t, 24 h. (f) *m*-CPBA (2 eq.), DCM, 0°C \rightarrow r.t., 24 h.

Synthesis of 6-amide-benzothiazoles **4c-r** was performed as depicted in **Scheme 3**. Commercially available 2-amino-6-carboxybenzothiazole was reacted with the appropriate amine **9c-f**, **i-j**, **l-r** through standard coupling conditions with EDC and HOBt as condensing agents, in DMF at room temperature for 24 hours. Amines **9c-f**, **i-j**, **l-o** were commercially available and directly used.



Scheme 3. Reagents and conditions: (a) EDC (1 eq.), HOBt (1 eq.), DMF, 0°C to r.t, 24 h; (b) 1N NaOH aq. (1.5 eq.), THF:MeOH 3:1, r.t., 24 h. (c) methyl isonipecotate (1 eq.) or benzylamine (1 eq.), EDC (1 eq.), HOBt (1 eq.), DMF, 0°C to r.t, 24 h; (d) Zn (2 eq.), NH₄Cl (2.5 eq.), MeOH, refl. 6 h.

Contrary, amines **9p-r** were synthesized first through $S_N 2$ reaction between the appropriate benzylamines and alkylbromides (**Scheme 4**).



 $\begin{array}{l} \textbf{9p:} R: \ Cl-; \ R^1: \ Cl-; \ R^2: \ C_6H_5-CH_2-\\ \textbf{9q:} \ R: \ H-; \ R^1; \ H-; \ R^2; \ HO-CH_2CH_2-\\ \textbf{9r:} \ R: \ Cl-; \ R^1: \ Cl-; \ R^2: \ 4-(CH_3O-)C_6H_2-)CH_2-\\ \end{array}$

Scheme 4. Reagents and conditions: (a) benzylamine (2.5 eq.), alkylbromide (1.0 eq.), K₂CO₃ (2.5 eq.), DMF, r.t., 1-6 h.

6-amido-4-hydroxy-benzothiazoles **4s-t** were prepared as depicted in Scheme 5. 2-amino-6hydroxy-benzothiazol-6-carboxylic acid **10** was synthesized first by cyclization of the respective aniline with potassium thiocyanate and bromine, in glacial acetic acid as previously described. **10** was further condensed with benzylamine (to give **4s**) or 3,4-dichlorobenzylamine (to give **4t**) in standard coupling conditions (**Scheme 5**).



Scheme 5. Reagents and conditions: (a) Br_2 (2 eq.), KSCN (4 eq.), glacial acetic acid, T < 10 °C, 1-6 h. (b) benzyl amines (1 eq.), EDC (1 eq.), HOBt (1 eq.), DMF, 0°C to r.t, 24 h.

Finally, for the synthesis of the catecholic 4-amido-benzothiazoles **5a-c**, the benzothiazole core was synthesized first starting from commercially available methyl 2-amino-4,5-dimethoxybenzoate by reaction with potassium thiocyanate and bromine, in glacial acetic acid as previously describe. The intermediate **11** was hydrolyzed to carboxylic acid **12** with aqueous NaOH and further condensed with the appropriate benzyl amine by coupling reaction. The methoxy groups of intermediates **13a-c** were deprotected with BBr₃ in DCM under nitrogen atmosphere to give the final products **5a-c** (**Scheme 6**).



Scheme 6. Reagents and conditions: (a) Br_2 (2 eq.), KSCN (4 eq.), glacial acetic acid, T < 10 °C, 1-6 h. (b) 1N NaOH aq. (1.5 eq.), THF:MeOH 3:1, r.t., 24 h. (c) benzyl amines (1 eq.), EDC (1 eq.), HOBt (1 eq.), DMF, 0°C to r.t, 24 h. (d) BBr₃ 1M in DCM (6 eq.), dry DCM, N₂, 0 °C to r.t., overnight.

Relationships between the structural properties, the enzymatic activity and the binding mode to *Tb*PTR1.

All 42 new benzothiazoles 1a-g, 2a-h, 3a,c-g, 4c-t and 5a-c were evaluated for their inhibitory activity against T. brucei and L. major PTR1, initially at 50 μ M and those compounds yielding > 60 % inhibition were profiled in dose-response experiments to determine their potency. The results are reported in Tables 1-5 and pyrimethamine was routinely utilised as a positive control compound (TbPTR1 IC₅₀ 0.09 µM and LmPTR1 IC₅₀ 13.6 µM). Almost all the synthesized compounds showed a measurable IC₅₀ (< 100 μ M) in the range of 0.3 – 57.6 μ M against *Tb*PTR1 and 1.9 – 94 μ M against LmPTR1. Eight compounds against TbPTR1 (1a, 2a, 3d-f, 4i,s-t) and 17 compounds against *Lm*PTR1 (1a, c-d, 2a, 3c-f, 4g, i, m-r, t) yielded IC₅₀ > 100 μ M and therefore were considered inactive. Overall, the compounds resulted from 2 to 100-fold more active against TbPTR1 than LmPTR1. Only compounds 3a, 4j and 5a-c showed comparable activity against both PTR1 enzymes, whereas the only inversion in tendency was observed for 4s showing LmPTR1 IC₅₀ = 2.48 µM but no inhibitory activity against *Tb*PTR1. The entire structure activity relationship is resumed in Figure 6. In addition, the LE against *Tb*PTR1 was calculated for all the synthesized compounds. LE, instead of IC₅₀, removes the bias due to the increased size of the ligands, highlighting the structural modifications favourable or deleterious for the inhibitor. Focusing on TbPTR1, the elongation of the 2-amino-benzothiazole main scaffold in a folic acid substrate-like shape resulted in a worthwhile strategy. The introduction of an aromatic ring as a side chain, able to mimic the pABA ring of folic acid, led to low/sub micromolar TbPTR1 inhibitors. Docking calculations predicted the side aromatic ring to be mainly placed in hydrophobic pocket 1, forming face-to-face and edge-to-face π - π interactions with Trp221 or Phe97, respectively. On the contrary, compounds 1-3a or 4m, possessing a methyl group or a non-aromatic ring, respectively as the side chain, were completely inactive. The only exceptions were compounds 1b and 2b possessing a CF₃ group as the side chain (*Tb*PTR1 IC₅₀ = 15.1 and 0.50 μ M for **1b** and **2b**, respectively). In particular, the

replacement of the oxygen atom of the starting benzothiazole **1b** with a sulphur atom in compound **2b** resulted in a 30-times improvement of the inhibitor activity against *Tb*PR1. The efficacy of this chemical modification is reflected in the meaningful improvement in LE for **2b** (LE = 0.42) compared to **1b** (LE = 0.32). The role of the linker was subsequently analysed. Sulfide derivatives (*Tb*PTR1 IC₅₀ = 0.35 – 1.9 μ M), followed by etheric derivatives (*Tb*PTR1 IC₅₀ = 0.67 – 15.1 μ M) resulted in the most active compounds of the library, compared to the homologous sulfone and amide derivatives. The efficacy and the importance of the linker is reflected in the overall range of LE of each subseries of compounds. Comparing benzothiazoles differing only for the linker between the main core scaffold and the side chain, the higher LE was always observed for sulphide (LE = 0.20 – 0.42) and etheric derivatives (LE = 0.18 to 0.32), followed by sulfones (LE = 0.14 – 0.25) and amides (LE = 0.15 – 0.26). The introduction of small polar substituents on the aromatic ring (e.g. nitrile, nitro, amino and methyl ester) was always tolerated (*Tb*PTR1 IC₅₀ **1e** = 0.67 μ M; **1f** = 1.7 μ M; **2e** = 0.35 μ M; **2f** = 0.90 μ M; **3e** = 3.00 μ M) resulting in a LE ranging between 0.25 (for **3e**) and 0.31 (for **1e**). On the contrary, bulky side chains (e.g. phenyl ring in **4l**, indole in **4o** or trimethoxy in **4i**) led to a drop both in activity and LE (< 0.19).

In our previous work¹⁶ we stated that the substitution of folate glutamate with the cyclic analogous methyl isonipecotate was well tolerated for inhibitory activity against PTR1s. This structural modification was therefore introduced on 2-amino-benzothiazole scaffolds to closely match the structures of folic acid, resulting in compounds **1-4g** and **2,4h**. The improvement in activity for these derivatives was only modest, as suggested in addition by a LE < 0.20, probably due to the exposition of the folic acid side chain to the solvent (*Tb*PTR1 IC₅₀ **1g** = 3.3 μ M; **2g** = 0.93 μ M; **3g** = 24.3 μ M; **4g** = 21.8 μ M).

The introduction of a hydroxyl group in position 4 on the benzothiazole ring (**4s-t**) resulted in a significant reduction in inhibitory activity against *Tb*PTR1. From docking calculation, the hydroxyl group appears to force the benzothiazole ring to assume an unusual pose within the active site. The

scaffold is flipped by 180° with the intra-ring sulphur pointing toward Tyr174, and the benzothiazole displaced from the standard stacking geometry between Phe97 and NADP(H).

All the tested compounds were twice to 100-times less active against *Lm*PTR1 than *Tb*PTR1 reflecting the efficacy of the design which was mainly performed on the architecture of the *Tb*PTR1 enzyme. Even though an overall comparable SAR among the two enzymes can be observed, these differences are presumable due to the structural diversity of the distal region of the binding site of the two enzymes. Contrary to *Tb*PTR1, *Lm*PTR1 possesses a more polar and solvent-exposed active site entrance surface.²⁴ Thus, the elongation of the 2-amino-benzothiazole main scaffold in a folic acid substrate-like shape, pointed the side chain toward a polar area, preventing the capability of the side benzyl ring to form π - π stacking or additional hydrophobic contacts (as in *Tb*PTR1), probably explaining the observed lower activity against *Lm*PTR1 compared to *Tb*PTR1.



Figure 6. Schematic highlights of the structure-activity relationship of the synthesized compounds against *Tb*PTR1 and *Lm*PTR1. All the inhibitors resulted more active against *Tb*PTR1 than *Lm*PTR1. The above reported structure-activity considerations are consistent for both enzymes.

A notable exception is represented by three compounds of series 5 (5a-c), showing comparable inhibitory activity against both enzymes. From docking calculations, these sub-classes of

benzothiazoles could assume a different binding mode (Figure 5D) compared to the poses of series 1-4, allowing them to adapt to both *Tb*PTR1 and *Lm*PTR1 binding sites.

Comparison between crystallographic ternary complexes of the inhibitors with TbPTR1 and -NADP(H) with the docked binding modes.

Compounds showing IC₅₀ values lower than 1 μ M and a low toxicity and seven compounds belonging to series 4 bearing a high structural diversity were subjected to structural studies. The structure of *Tb*PTR1 was solved in complex with the cofactor NADP(H) and nine benzothiazole derivatives (**1e**, **2b**, **2d**, **2g**, **2h**, **4c**, **4d**, **4g** and **4l**) showing IC₅₀ values ranging from 0.5 μ M (compounds **2b** and **2h**) to 21.8 μ M (compound **4g**). In all structures, described in Table 6, the shared benzothiazole core of the inhibitors adopts a highly conserved binding mode, retaining also the same interaction patterns described for the forefather **3a**.

Table 6. Crystal structures of *Tb*PTR1 in complex with NADP(H) and the benzothiazole compounds. A brief description of the main interactions entailed by the inhibitors in the catalytic pocket is reported (the shared benzothiazole moiety forms conserved interactions in the cavity, as already described for the forefather **3a**). The picture column reports an active site view of *Tb*PTR1 (white cartoon and carbon atoms) in complex with the cofactor NADP(H) (in sticks, black carbons) and **1e** (in sticks, green carbons), **2b** (in sticks, cyan carbons), **2d** (in sticks, light blue carbons), **2g** (in sticks, brown carbons), **2h** (in sticks, liac carbons), and **4e** (in sticks, orange carbons), **4d** (in sticks, dark green carbons), **4g** (in sticks, dark red carbons), and **4l** (in sticks, pink carbons). Inhibitors are surrounded by their omit map (green wire) contoured at the 2.5 σ level. Water molecules are represented as red spheres and H-bonds as red dashed lines. Atom color code: nitrogen (blue), oxygen (red), sulfur (yellow), phosphorous (magenta), chlorine (dark grey) and fluorine (grey).

Complex	Picture	Description
<i>Tb</i> PTR1 – NADP(H) – 1e	Cys168 Phe171 Trp221 1e Asp161 Ser95 Met213 Pro210 NADP+	 PDB: 6GCK Resolution: 2.14 Å 4-cianophenyl moiety of 1e (IC₅₀ of 0.67 μM) is stabilized in pocket 1 by van der Waals interactions with Phe97, Leu209, Pro210, Met213, and Trp221 (interactions shared within the tetramer). 1e populates four subunits of the tetramer (100 % occupancy).







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The analysis of the ten ternary X-ray complexes (Figure 2 and Table 6) revealed a conserved binding mode and some common interactions of the ten inhibitors within *Tb*PTR1. In all ten complexes, the benzothiazole core is always located within the catalytic pocket of *Tb*PTR1, interacting with the four key amino acids (Ser95, Phe97, Asp161 and Trp221) and NADP(H), as previously described (Figure 7). The main contribution to binding originates from the π -stacking between the nicotinamide moiety of NADP(H) and Phe97 and H-bonds with phosphate or ribose hydroxyls of the cofactor. An additional conserved H-bond interaction is always observed between the amino group in position 2 on the benzothiazole ring and the hydroxyl group of Ser95. Additional weak electrostatic interaction with Asp161 was observed for sulfone (for **3a**) or amide (for **4d**,**g**,**I**) benzothiazoles. Moreover, for all the inhibitors possessing an aromatic ring as the side chain, mimicking the *p*ABA moiety of the folate substrate (**1e**, **2d**,**g**,**h**, **4g**,**l**), the side-chain is always located within the hydrophobic pocket 1 stabilized by van-der-Waals (vdW) interactions with Leu209, Pro210, Met213, and Trp221, and mainly by face-to-edge π - π stacking with Phe97 (Figure 7).



Figure 7. Common binding interactions of benzothiazoles with the main amino acids delimiting the *Tb*PTR1 active site and NADPH based on the crystallographic ternary complex of compounds **1e**, **2b**, **d**, **g**-**h**, **3a**, **4c**-**d**, **g**, **l**.

The binding mode of the ten benzothiazoles crystalized was compared by superimposition with the respective docking poses (Figure 8B-K). In both X-ray and docked poses, the main 2-aminobenzothiazole core always shows the same binding mode and orientation within the *Tb*PTR1 catalytic pocket. In addition, the orientation and placement of the side chain are well reproduced for the inhibitors with small 'tails' attached to the main benzothiazole scaffold (such as for compounds **1e**, **2b**,**d**, **3a**, **4c**,**d**,**l**). On the contrary, for compounds **4g** (Figure 8D), **2g** (Figure 8J) and **2h** (Figure 8K) the distal part of the longer side chain is not well reproduced, since it points outside the confined *Tb*PTR1 pocket and can assume multiple conformations. An overall good correlation between the *Tb*PTR1 pIC₅₀ of all inhibitors and the respective Autodock Docking Score was observed (Figure 8A, Table SI-4), revealing the capability of the docking protocol and the design and selection strategy adopted to foresee valid potential *Tb*PTR1 inhibitors.



Figure 8. (A) Correlation between the enzymatic inhibitor activity against *Tb*PTR1 (expressed as pIC₅₀) and the Autodock Score. Colored dots correspond to the inhibitors for which the X-ray complex is available. The color of the dots is in accordance with the chemical series: ether (blue), sulfide (yellow), sulfone (red), amide (green). (B-K) Superimposition between the binding mode observed by X-ray crystallography and the respective pose predicted by docking calculation for inhibitors **4c** (B, carbon in light orange for X-ray pose and in red for docked pose), **4d** (C, carbon in light green for X-ray pose and in dark green for docked pose), **4g** (D, carbon in yellow for X-ray pose and in orange for docked pose), **4l** (E, carbon in purple for X-ray pose and in violet for docked pose), **3a** (F, carbon in purple for X-ray pose and in violet for docked pose), **1e** (G, carbon in light green for X-ray pose and in dark green for docked pose), **2b** (H, carbon in cerulean for X-ray pose and in blue for docked pose), **2d** (I, carbon in cerulean for X-ray pose and in blue for docked pose), **2g** (J, carbon in purple for X-ray pose and in violet for docked pose), **2g** (J, carbon in purple for X-ray pose and in violet for docked pose), **2h** (K, carbon in yellow for X-ray pose and in orange for docked pose). The X-ray poses are colored in lighter tone and the docked pose in a darker tone. Color code: nitrogen (blue), oxygen (red), sulfur (yellow), phosphorous (magenta), chlorine (dark green), fluorine (green).

Anti-parasitic activity as single agents and in combination

All synthesized benzothiazoles were evaluated first for *in-vitro* anti-parasitic activity against cultured bloodstream forms of T. brucei. To evaluate a potential broad-spectrum antitrypanosomatidic activity all the compounds were in addition screened against the amastigote form of Leishmania infantum (L. infantum) and trypomastigote form of T. cruzi, which reproduce a more physiological and disease-relevant model. Human leukemic monocyte cell line THP-1 was infected with L. infantum while osteosarcoma human U2OS cell-line was infected with trypomastigote forms of the Y strain of T. cruzi. Compounds were screened at 10 µM against T. brucei and intracellular amastigote L. infantum, and at 50 µM against intracellular amastigote T. cruzi during a HTS phenotypic screening campaign that included thousands of compounds. The anti-parasitic activity was expressed as % of parasitic cell growth inhibition at the defined compound concentration. For compounds associated with > 60 % cell growth inhibition (against *T. brucei* and L. infantum) and > 85 % cell growth inhibition (against T. cruzi), dose-response studies were performed to determine the EC₅₀ values (Table 1-5). The T. brucei assay relied on indirect determination of parasite population viability by quantification of total DNA present in the well using the Sybr Green I DNA fluorescent dye.³² Pentamidine was used as the reference compound exhibiting an EC₅₀ of 3.8 nM which is comparable with the value reported in literature.³³ Amphotericin B and miltefosine were utilized as positive controls in the L. infantum assay yielding EC₅₀ 1.14 µM and 1.29 µM respectively, while benznidazole was used as reference compound in the T. cruzi assay, exhibiting an EC₅₀ of 2.4 μ M, which is comparable with the value reported in literature.³⁴ Almost all tested benzothiazoles yielded weak anti-T. brucei activity ranging below 20 % cell growth inhibition at 10 μ M (Table 1-5). Only three compounds (4c, 4r and 4t) showed a % cell growth inhibition > 60 % at 10 μ M resulting in an EC₅₀ of 7.0 μ M (4c), 9.7 μ M (4r) and 5.9 µM (4t) respectively. In contrast, no compounds showed an appreciable anti-parasitic activity

against the two intracellular forms of *L*. *infantum* and *T*. *cruzi* (% cell growth inhibition < 21 % at 10 μ M, Table 1-5).

As stated before, the effective inhibition of parasite growth can be observed if both PTR1 and DHFR enzymes are inhibited.¹⁷ On this basis, we studied the effect of the PTR1 inhibitors in combination with MTX, a known DHFR inhibitor, against *T. brucei* to observe the potentiation effect that the PTR1 inhibitors have on MTX parasite growth inhibition. With this aim, we selected some inhibitors showing the highest inhibition effect against *Tb*PTR1, namely, compounds **1d-e**, **2b,d-h**, **3e** and **4c**. The selected compounds were combined at 10 μ M with MTX at 4 μ M (corresponding to the EC₃₀ against *T. brucei*).²⁴ Only compound **4c** was tested at 1 μ M, because it showed an anti-parasitic activity >80% at 10 μ M which is an inhibition potency too high to allow any potentiation effects in combination to be observed. The anti-parasitic activity was expressed as % cell growth inhibition and is reported in Figure 9 and Table 7. The potency in combination studies was estimated through determination of a Potentiating Index (PI) given by **eq. 1**.

$$PI = \frac{\% \text{ inhibition of the combination}}{(\% \text{ inhibition of compound } + \% \text{ inhibition at } 4 \,\mu\text{M MTX})}$$
eq. 1

A PI higher than 1 reflects the capacity of the compounds to increase the anti-parasitic activity of MTX beyond what would be expected by the simple addition of the individual effects (Figure 9 and Table 7). A PI effect was observed for all compounds except for 2d and 2b. Compounds 1e and 2f were most potent compounds showing PI >2.

Table 7. *T. brucei* antiparasitic activity of compounds **1d-e**, **2b,d-h**, **3e**, **4c** tested as single agents (at 10 μM) and potentiation of MTX activity in combination.

Compound	% T. brucei growth inhibition		Ы
-	Single agent (at 10 µM)	+ 4 μM MTX	11
MTX	19 ± 6	Not detectable	-
1d	55 ± 1	89 ± 8	1.2 ± 0.1
1e	21 ± 5	80 ± 2	2.0 ± 0.0
2b	11 ± 0	33 ± 7	1.1 ± 0.1
2d	41 ± 4	65 ± 4	1.1 ± 0.0
2e	29 ± 5	63 ± 1	1.3 ± 0.3
2f	7 ± 11	69 ± 7	2.7 ± 0.4
2g	53 ± 3	93 ± 1	1.3 ± 0.0
2h	33 ± 12	80 ± 0	1.6 ± 0.2
3 e	13 ± 4	48 ± 8	1.5 ± 0.2
4c ^a	Not detectable	30 ± 5	1.6 ± 0.1

a. tested at 1 μM



Figure 9. Anti-parasitic activity against *T. brucei* of compounds 1d-e, 2b,d-h, 3e tested as single agent at 10 μ M (in cerulean) and in combination with 4 μ M of MTX (in vermillion). Compound 4c was tested at 1 μ M as a single agent. The anti-parasitic activity of MTX as single agent at 4 μ M is reported as a blue bar. As a quantitative measure of potentiation, a PI of the combination was determined (lined bar).

Early ADME-Toxicity studies

Early toxicity studies were performed for all compounds synthesized with the aim to identify liabilities and to select the safest compounds among the most active compounds, for progression towards further studies. The panel of early toxicity assays included (1) cardiac toxicity by measuring the inhibitory effect on human cardiac potassium channel (*h*ERG), (2) inhibition of five major drug metabolizing cytochrome P450 enzymes (CYP 1A2, 2C9, 2C19, 2D6, 3A4), (3) cytotoxicity against three human cell-line, namely A549 (human lung adenocarcinoma epithelial cells), THP-1 (macrophages-like cells), U2OS (osteosarcoma cells) and (4) mitochondrial toxicity in 786-O (renal carcinoma) cell-line. Compounds were initially screened at 10 μ M against *h*ERG, CYPs, mitochondria and A549 cell-lines and at 50 μ M against U2SO cells and for active compounds, the respective IC₅₀ values and GI₅₀ against cell lines were experimentally determined. Cytotoxicity against THP-1 macrophages was expressed as No Observed Adverse Effect Level (NOAEL), measuring the maximum dose administered without observing toxic effects. The early ADME-toxicity studies were performed for all the synthesized compounds **1a-g**, **2a-h**, **3a,c-g**, **4c-t**, **5a-c** and reference drugs pentamidine. The data were organized using a traffic light representation for rapid and intuitive visualization of the complex data set (Figure 10).

To reduce the liabilities of the compounds an IC₅₀ or GI₅₀ > 10 μ M was set as the cut-off for an acceptable liability profile²⁴, following the features a compound that should behave as a good inhibitor *in vitro*, should have for progression to animal studies according to the TIP. An ideal compound should have all parameters colored in green in Figure 9. Almost all the evaluated benzothiazoles showed a safe early toxicity profile against *h*ERG, CYP isoforms and cell-lines. Only a few compounds showed toxicity (IC₅₀ < 10 μ M) against *h*ERG (i.e. **1a**, **1c**, **2a**, **4g**, **4n** and **5a**) and mitochondria (i.e. **1a**, **2a**, **2e**, **4d**, **4g** and **4j**). No cytotoxicity was observed against U2OS and THP-1 and only three compounds resulted cytotoxic against A549 with GI₅₀ < 10 μ M (**1a**, **2b** and **2h**). In contrast, benzothiazoles appeared to interfere with the activity of CYP isoforms with

particular impact on CYP2D6 and CYP3A4. However, the absolute IC_{50} and GI_{50} should be compared with the EC_{50} against the target parasite for the determination of a Selectivity Index (SI), calculated as a ratio of the IC_{50} or GI_{50} against the liability target and the anti-parasitic EC_{50} . This was used for the qualitative evaluation of the dose of compound necessary to achieve the desired pharmacological effect and minimal toxicity. Accordingly to our TIP, a SI > 10-fold between the anti-parasitic activity and the liability targets and/or cytotoxicity was preferred in order to consider the compounds safe for *in vivo* evaluation. Thus, **4c**, **4r** and **4t** which exerted anti-parasitic activity against *T. brucei* as single agents (**4c** $TbEC_{50}$ 7.0 ± 0.5 μ M; **4r** $TbEC_{50}$ 9.7 ± 2.5 μ M; **4t** $TbEC_{50}$ 5.9 ± 0.2 μ M) enabled calculation of their SI values, which are reported in Figure 11. For compounds associated with IC_{50} , GI_{50} and NOAEL > 100 μ M, the respective SI is underestimated. From these studies, **4c** showed a suitable selectivity (SI > 10) with respect to its anti-*T. brucei* activity, and an acceptable early toxicity profile that fits the expected TIP criterion⁴⁰ for progression of promising hits to *in vivo* pharmacokinetic studies.



Figure 10. Early toxicity profile of benzothiazoles evaluated in the present studio and the reference drug pentamidine. The data were expressed in term of micromolar IC₅₀ against *h*ERG, CYP isoforms and mitochondrial toxicity, micromolar GI₅₀ against A549 and U2OS cell lines and NOAEL (maximum dose (μ M) administered without observing toxic effects). The cells are colored adopting a trafficlight system in vermillion for high toxic potential (IC₅₀ or GI₅₀ or NOAEL < 10 μ M), in white for moderate toxicity (IC₅₀ or GI₅₀ or NOAEL ranging from 10 – 25 μ M) and in cerulean for low/no toxicity (IC₅₀ or GI₅₀ or NOAEL > 10 μ M). To reduce the liabilities of the compounds (acceptable liability profile) an IC₅₀ or GI₅₀ or NOAEL > 10 μ M (cerulean/white color) was set as a cut-off.


Figure 11. Early toxicity profile (IC₅₀ against *h*ERG, CYPs and mitochondria, GI₅₀ against A549 and U2OS and NOAEL against THP-1) of compounds **4c** (dark blue), **4r** (blue) and **4t** (cerulean) and relative selectivity index (SI, in dark red, vermillion and pink, respectively) with respect to *T*. *brucei* anti-parasitic activity (EC₅₀).

Pharmacokinetic evaluation of 4c

Before proceeding to *in vivo* studies, the solubility of **4c** was evaluated first by UV-Vis spectroscopy as describe by Bard et al.³⁵ The thermodynamic solubility was measured in four different media (*i.e.* pure water, phosphate buffered saline (PBS), PBS + 10 % DMSO and PBS + 50 % DMSO) in accordance with the administration procedure. Maximum solubility in water and PBS was 13.12 µg/mL and 10.94 µg/mL respectively, and > 100 µg/mL in both PBS with 10 % or 50 % of DMSO, resulting a moderately-highly soluble compound for *in vivo* evaluation.³⁶ The preliminary pharmacokinetic profile of **4c** was evaluated in healthy BALB/c mice using a SNAP-PK approach and the free form of **4c** was administered IV at the dose of 1 mg/kg and *per os* at the dose of 20 mg/kg (Table 8). Compound **4c** showed a short half-life ($t_{1/2} = 3.45$ h) after IV administration and it was not detected in blood after 2 hours. In addition, neither after IV nor *per os*

administration the compound reached a plasmatic concentration comparable or higher to its anti-*T*. *brucei* activity (EC₅₀ = 7.0 μ M). With the aim of improving the plasma concentrations, **4c** was solubilized with hydroxypropyl- β -cyclodextrin (HP- β -CD) and administered *per os* at the dose of 20 mg/kg. Formulated with HP- β -CD, compound **4c** reached a maximum plasmatic concentration (C_{max}) of 13 μ M (3545 ng/mL), 1.9 times higher than the respective *T. brucei* EC₅₀. The formulation **4c** + **CD** showed, in addition, an improved half-life (t_{1/2} = 34.8 h) keeping the plasmatic concentration higher than the anti-parasitic EC₅₀ for the first hour after administration (Table 8).

Table 8. Pharmacokinetic parameters of 4c administered IV and per os

4 c	Dose	Cmax	AUC^a	t _{1/2} (h)	Clearence
		(ng/mL)	(ng/mL∙h)		(mL/min/kg)
IV	1 mg/kg	1630	1057	3.45	946
per os	20 mg/kg	826	4686	43.5	68
per os + CD	20 mg/kg	3545	7932	34.8	41

a. Area under the curve

The pharmacokinetic profile could be modulated and improved by the formulation with HP- β -CD (Table 8). **4c** was then progressed *in vivo* against *T. brucei* infected BALB/c mice. The observed pharmacokinetic improvement in healthy mice was insufficient to enable the significant reduction of the infection or increase the survival of the animals with respect to the untreated control (data not shown). The potential effect of the compound will be further explored going back to chemical modification and different delivery systems.

CONCLUSION

The aim of the present work was to enhance the benzothiazole library through a structure-based approach, to characterize the biological activity against PTR1 and Trypanosomatidic parasites and evaluate the ADME-Toxicity profile to select at least one compound for progression to pharmacokinetic evaluation in animal assays.

We were able to expand the library around the previously identified riluzole and a few other benzothiazole derivatives. The initial X-ray crystal structure of TbPTR1 with the compound of reference, 2-amino-benzothiazole, guided the structure-based design of the new library. We designed and synthesized 42 new 2-amino-benzothiazoles that were investigated for their on-target activity (both T. brucei and L. major PTR1), anti-parasitic effect, and early toxicity in vitro. These experiments allowed the compounds' biological profile to be largely explored, not only a specific description of the targets' inhibition such as two PTR1 enzymes and the corresponding parasitic DHFR other than the corresponding human enzymes, but also the effect against different other enzymes and cells involved in compounds toxicity. Among them, at least six CYP enzymes inhibition, the *h*ERG channel binding, the mitochondrial inhibition effects and different human cells including the human macrophages THP-1. The huge amount of data obtained represents an added value to the compound characterization in a very peculiar way; in fact, while the biological effect on the parasitic enzymes and cells are specifically directed to assess the potential anti-parasitic activity, the ADME-Toxicity data represents an independent dataset that can help in the understanding the crucial structural fragment responsible for the compounds toxicity, other than provide data that are definitely associated with the compounds as their chemical-physical properties, their structural features and will be available for further research even in other medicinal chemistry fields.

Ten new x-ray crystal structures of PTR1/compound complexes were solved, and the crystallographic binding mode observed reproduced the poses predicted by the docking studies, validating and confirming the robustness of the *in-silico* approach adopted. It was possible to explain how the compounds structural features modulate the compounds interactions with the targets.

Progression of compounds into *in vivo* evaluation was based on a compound's property profile its efficacy and safety compared to the features that the expected ideal compound should have.³⁷ We

therefore defined the profile of the selected compounds and the criterion for go/no go in the drug discovery process as reported in Figure 12. The criteria included chemical (*i.e.* purity, undesirable chemical function, synthetic cost and solubility) and molecular biology properties (i.e. knowledge of the target, crystal structures, activity against the target enzyme and SAR), anti-parasitic activity against the three parasites being objects of this study, early enzymatic, cytotoxicity and *in vivo* pharmacokinetic parameters (Figure 12, Table SI-5).⁴⁰ Pentamidine, the first line drug for the treatment of HAT was used as a reference compound and its chemical, biochemical, anti-parasitic, toxicity and pharmacokinetic profile has been reported³⁸ and serves as a guide to benchmark the TIP and identify compounds for progression. Among all the 42 new benzothiazoles evaluated in the present work, **4c** was the most promising compound and fitted the progression criteria summarized in Table SI-5, for each property shown in Figure 12, as it inhibited the target enzymes, had anti-parasitic activity and an acceptable early toxicity profile (enzymatic liability and cytotoxicity). Compound **4c** and pentamidine profiles are reported in Figure 12.



Figure 12. TIP of benzothiazole **4c** (blue line) compared with the reference drug pentamidine (red line). For parameters requiring a yes/no answer, the cut-off was set at 50% of the y-axes. For

parameters requiring a numeric value, the marker line between proper and improper behaviour was set to the corresponding cut-off value described in the main text. Pale green colour covers the area for which the compound parameters agree with the desired values, whereas pale red colour covers the area with an improper profile. Red dots represent data relative to pentamidine, blue dots are related to **4c**.

Despite the apparent higher toxicity of pentamidine compared to **4c** (some dots related to toxicity are in the red field, Figure 12), the drugs higher potency favors a higher specificity in vitro. Significantly, **4c** presented oral availability, in contrast to pentamidine, a major hallmark in drug development for these diseases.

Although **4c** was optimal for trypanotoxicity, inhibition of *Tb*PTR1 *in vitro*, had a promising early toxicity and pharmacokinetic profile for oral administration it was not curative in the *in vivo* model.

Despite the *in vivo* result, this class of compounds offer the potential for optimization into novel oral drugs for the treatment of HAT. The detailed chemical and biological information obtained from these studies provides a rational for the design of further series of compounds with improved PTR1 enzyme inhibition and anti-parasitic activity. Future work will address the formulation of **4c** in a more suitable stable delivery system that could enhance its *in vivo* anti-parasitic activity.

EXPERIMENTAL SECTION

Synthetic procedures. All commercially available chemicals and solvents were reagent grade and were used without further purification unless otherwise specified. The following solvents and reagents have been abbreviated: methanol (MeOH); acetonitrile (ACN); ethyl acetate (AcOEt); formic acid (FA); 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC); Hydroxybenzotriazole (HOBt). Reactions were monitored by thin-layer chromatography on silica gel plates (60F-254, E.

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Merck) and visualized with UV light, cerium ammonium sulfate or alkaline KMnO₄ aqueous solution. NMR spectra were recorded on a Bruker 400 spectrometer with ¹H at 400.134 MHz and ¹³C at 100.62 MHz. Proton chemical shifts were referenced to the solvent residual peaks. Chemical shifts are reported in parts per million (ppm, δ units). Coupling constants are reported in Hertz (Hz). Splitting patterns are designed as s, singlet; d, doublet; t, triplet; q quartet; dd, double doublet; m, multiplet; b, broad. Melting points were recorded on a Stuart, SMP3 (Barloworld Scientific Limited Stone, Staffordshire, UK) and are uncorrected. Analysis of compound purity was determined through liquid chromatography (LC) UV/Vis using a Jasco LC system equipped with a Jasco PU-2080 Plus pump, coupled with a Jasco PU-2075 Plus UV/Vis detector. LC separation was performed on an Agilent Poroshell 120 50 mm × 3.0 mm analytical column, packed with EC-C18 2.7 μ M as stationary phase (Agilent Technologies, Milan, Italy). 20 μ L of a 100 μ g/mL solution of compound in 0.1% FA in water/ACN 9:1 was injected. A gradient was delivered at 0.2 mL/min using (A) 0.1% FA in water and (B) ACN. Samples were eluted with 5% B (0.00–1.00 min); 1–95% B (1.00–6.00 min); 95% B (6.00-10.00 min) and 5% B (10.00–20.00 min). The eluate was detected by UV at $\lambda = 220$ nm. All the compounds showed a level of purity above 95%.

General procedure for the synthesis of benzothiazoles 1a, c-f

To a solution of **6** (1 eq.) in DMF (5 mL), K_2CO_3 (2.5 eq.) and iodomethane (1.1 eq., for **1a**) or the appropriate benzylbromide (1.2 eq., for **1b-f**) were added. The mixture was stirred at room temperature overnight. The solvent was evaporated under reduced pressure and the residue suspended in AcOEt and washed with K_2CO_3 saturated solution and brine. The organic phase was dried over Na₂SO₄ and concentrated. The crude was chromatographed on silica gel to afford the desired product.

6-methoxybenzo[d]thiazol-2-amine (1a)

White solid. 64% yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.74 (s, 3H), 6.81 (dd, J = 2.7, 8.7 Hz, 1H), 7.14 – 7.28 (m, 3H), 7.29 (d, J = 2.6 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 55.50, 105.49, 112.82, 118.02, 131.87, 146.80, 154.23, 164.66. m.p. [164.2-165.8°C]. k' = 5.32. HRMS m/z [M+H]⁺ Calcd for C₈H₈N₂OS: 180.0357. Found: 180.0360.

6-(benzyloxy)benzo[d]thiazol-2-amine (1c)

White solid. 84% yield. ¹H NMR (400 MHz, DMSO- d_6) δ 5.06 (s, 2H), 6.87 (dd, J = 2.6, 8.7 Hz, 1H), 7.14 – 7.27 (m, 3H), 7.26 – 7.55 (m, 6H). ¹³C NMR (101 MHz, DMSO) δ 69.80, 106.73, 113.68, 118.01, 127.64, 127.71, 128.35, 131.81, 137.26, 147.02, 153.24, 164.78. m.p. [135°C with dec.]. k' = 7.23. HRMS m/z [M+H]⁺ Calcd for C₁₄H₁₂N₂OS: 256.0670. Found: 256.0672.

6-((3,4-dichlorobenzyl)oxy)benzo[d]thiazol-2-amine (1d)

Pale yellow solid. 59 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 5.10 (s, 2H), 6.90 (dd, *J* = 2.6, 8.7 Hz, 1H), 7.24 (d, *J* = 8.5 Hz, 3H), 7.38 (d, *J* = 2.6 Hz, 1H), 7.44 (dd, *J* = 2.0, 8.3 Hz, 1H), 7.66 (d, *J* = 8.2 Hz, 1H), 7.71 (d, *J* = 1.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 68.26, 106.90, 113.67, 118.03, 127.74, 129.37, 130.21, 130.61, 131.02, 131.86, 138.57, 147.27, 152.80, 164.92. m.p. [174 °C with dec]. *k*' = 8.30. HRMS m/z [M+H]⁺ Calcd. for C₁₄H₁₀C₁₂N₂OS: 323.9891. Found: 323.9885.

6-(((2-aminobenzo[d]thiazol-6-yl)oxy)methyl)benzonitrile (1e)

Pale yellow solid. 69 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 5.20 (s, 2H), 6.91 (dd, *J* = 2.7, 8.7 Hz, 1H), 7.18 – 7.34 (m, 3H), 7.38 (d, *J* = 2.7 Hz, 1H), 7.66 (dd, *J* = 8.2, 10.0 Hz, 2H), 7.78 – 7.96 (m, 2H). ¹³C NMR (101 MHz, DMSO) δ 68.86, 106.85, 110.35, 113.65, 118.04, 118.73, 127.87, 128.01, 131.86, 132.35, 132.41, 143.16, 147.27, 152.81, 164.92. m.p. [196.2 – 197.0 °C]. *k*' = 6.25. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₁N₃OS: 281.0623. Found: 281.0625.

methyl 4-(((2-aminobenzo[d]thiazol-6-yl)oxy)methyl)benzoate (1f)

White solid. 85 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.86 (s, 3H), 5.18 (s, 2H), 6.91 (dd, J = 2.6, 8.7 Hz, 1H), 7.16 – 7.30 (m, 3H), 7.39 (d, J = 2.6 Hz, 1H), 7.59 (d, J = 8.2 Hz, 2H), 7.91 – 8.08

(m, 2H). ¹³C NMR (101 MHz, DMSO) δ 52.09, 69.17, 106.83, 113.68, 118.03, 127.42, 128.86, 129.25, 131.85, 142.88, 147.21, 152.99, 164.88, 165.98. m.p. [195 °C with dec.]. k' = 6.78. HRMS m/z [M+H]⁺ Calcd. for C₁₆H₁₄N₂O₃S: 314.0725. Found: 314.0725.

methyl 1-(4-(((2-aminobenzo[d]thiazol-6-yl)oxy)methyl)benzoyl)piperidine-4-carboxylate (1g)

1f (200 mg, 1 eq., 0.64 mmol) was solubilized in THF:MeOH 3:1 (8 mL) and 1N NaOH aqueous solution (960 μ L, 1.5 eq., 0.96 mmol) was added. The mixture was stirred at room temperature for 24 hours and then concentrated. The crude was solubilized in water and washed with AcOEt. The aqueous phase was acidified with 1N HCl and the precipitate formed, collected through filtration, washed with water and dried. The carboxylic acid (172 mg, 1 eq., 0.57 mmol) thus obtained was solubilized in DMF (5 mL) at 0 °C and EDC HCl (109 mg, 1 eq. 0.57 mmol), HOBt (76 mg, 1 eq., 0.57 mmol) and methyl isonipecotate (88 µL, 1 eq., 0.57 mmol) were subsequently added. The mixture was stirred at room temperature overnight and concentrated. The residue was suspended in AcOEt and washed with K₂CO₃ saturated solution and brine. The organic phase was dried over Na₂SO₄ and concentrated. The residue was crystalized from diethyl ether to give 145 mg of a white solid (61 % yield). ¹H NMR (400 MHz, DMSO-d₆) δ 1.51-1.54 (m, 2H), 1.75-1.98 (m, 2H), 2.67 (tt, J = 4.0, 11.0 Hz, 1H), 2.84 - 3.23 (m, 2H), 3.48-3.62 (m, 4H), 4.21-4.44 (m, 1H), 5.12 (s, 2H),6.80 - 7.01 (m, 1H), 7.17 - 7.31 (m, 2H), 7.38-7.43 (m, 3H), 7.50-7.54 (m, 3H). ¹³C NMR (101 MHz, DMSO) & 27.61, 40.48 (from HSQC), 44.44 (from HSQC), 51.52, 69.39, 106.74, 113.63, 118.03, 126.82, 127.48, 131.86, 135.52, 138.59, 147.12, 153.16, 164.84, 168.78, 174.21. m.p. $[127^{\circ}C \text{ with dec.}]$. k' = 7.42. HRMS m/z $[M+H]^+$ Calcd. for $C_{22}H_{23}N_3O_4S$: 425.1409. Found: 425.4010.

General procedure for the synthesis of sulfide-benzothiazoles 2a-f

To a suspension of aminobenzene-sulphides **8a-f** (1 eq.) in glacial acetic acid (10 mL *per* mmol) cooled at 10 °C, KSCN (4 eq.) was added. A solution of elemental bromine (2 eq.) in glacial acetic acid (10 mL *per* mmol) was added drop wise over 1 hour keeping the mixture refrigerated at a 40

temperature below 10 °C. After the addition, the temperature was spontaneously risen, and the mixture reacted at room temperature for 1 - 6 hours. The mixture was chilled down, quenched with water and alkalinized with 30% aqueous NH₄OH. The precipitated formed was extracted in AcOEt. The organic phase was washed with brine, dried over anhydrous Na₂SO₄ and concentrated. The crude was crystalized from diethyl ether to afford the desired product.

6-(methylthio)benzo[d]thiazol-2-amine (2a)

Pale yellow solid. 58 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 2.46 (s, 3H), 7.16 (dd, *J* = 2.0, 8.3 Hz, 1H), 7.26 (d, *J* = 8.3 Hz, 1H), 7.45 (s, 2H), 7.64 (d, *J* = 1.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 16.68, 117.92, 119.84, 125.51, 128.96, 132.06, 151.01, 166.22. m.p. [148.5 – 149.8°C]. *k*' = 5.72. HRMS m/z [M+H]⁺ Calcd. for C₈H₈N₂S₂: 196.0129. Found: 196.0130.

6-((trifluoromethyl)thio)benzo[d]thiazol-2-amine (2b)

Pale yellow solid. 32 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.41 (d, *J* = 8.4 Hz, 1H), 7.50 (dd, *J* = 2.0, 8.3 Hz, 1H), 7.88 (s, 2H), 8.07 (d, *J* = 1.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 112.75 (q, *J* = 3.0 Hz, 1C), 118.29, 129.28, 129.64 (q, *J* = 309 Hz, 1C), 132.29, 134.01, 155.39, 169.15. m.p. [154.2 - 156.1°C]. *k*' = 6.21. HRMS m/z [M+H]⁺ Calcd. for C₈H₅F₃N₂S₂: 249.9846. Found: 249.9845.

6-(benzylthio)benzo[d]thiazol-2-amine (2c)

Pale brown solid. 59 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.15 (s, 2H), 7.13 – 7.36 (m, 7H), 7.52 (s, 2H), 7.68 (dd, *J* = 0.5, 1.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 38.86 (overlapping with DMSO peaks), 117.82, 123.15, 126.28, 126.87, 128.23, 128.53, 128.76, 131.72, 137.90, 151.92, 166.76. m.p. [142.3 – 143.1°C]. *k*' = 7.45. HRMS m/z [M+H]⁺ Calcd. for C₁₄H₁₂N₂S₂: 272.0442. Found: 272.0445.

6-((3,4-dichlorobenzyl)thio)benzo[d]thiazol-2-amine (2d)

 White solid. 71 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.19 (s, 2H), 7.10 – 7.41 (m, 3H), 7.48 – 7.69 (m, 4H), 7.73 (d, J = 1.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 37.71, 117.85, 123.81, 125.12, 129.03, 129.09, 129.14, 129.33, 130.31, 130.64, 131.74, 139.58, 152.16, 166.99. m.p. [134.6 – 136.2°C]. k' = 8.50. HRMS m/z [M+H]⁺ Calcd. for C₁₄H₁₀(³⁵Cl)₂N₂S₂: 339.9662. Found: 339.9660. Calcd. for C₁₄H₁₀(³⁵Cl)(³⁷Cl)N₂S₂: 341.9633. Found: 341.9633.

4-(((2-aminobenzo[d]thiazol-6-yl)thio)methyl)benzonitrile (2e)

White solid. 65 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.22 (s, 2H), 7.15 (dd, J = 1.9, 8.3 Hz, 1H), 7.22 (d, J = 8.3 Hz, 1H), 7.36 – 7.48 (m, 2H), 7.55 (s, 2H), 7.66 (d, J = 1.9 Hz, 1H), 7.69 – 7.76 (m, 2H). ¹³C NMR (101 MHz, DMSO) δ 38.02 (overlapping with DMSO peaks), 109.53, 117.87, 118.75, 123.81, 125.04, 129.10, 129.71, 131.76, 132.12, 144.28, 152.24, 167.00. M.p. [164-7 – 166.0°C]. k' = 6.67. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₁N₃S₂: 297.0394. Found: 297.0395.

methyl 4-(((2-aminobenzo[d]thiazol-6-yl)thio)methyl)benzoate (2f)

White solid. 82 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.83 (s, 3H), 4.21 (s, 2H), 7.03 – 7.29 (m, 2H), 7.32 – 7.45 (m, 2H), 7.54 (s, 2H), 7.68 (d, J = 1.8 Hz, 1H), 7.80 – 7.89 (m, 2H). ¹³C NMR (101 MHz, DMSO) δ 38.77 (overlapping with DMSO peaks), 52.02, 117.84, 123.66, 125.43, 128.10, 128.99, 129.09 (overlapping of 2C signals), 131.73, 143.88, 152.17, 165.96, 166.91. m.p. [152°C with dec.]. k' = 7.20. HRMS m/z [M+H]⁺ Calcd. for C₁₆H₁₄N₂O₂S₂: 330.0497. Found: 330.0495.

Synthesis of methyl 1-(4-(((2-aminobenzo[d]thiazol-6-yl)thio)methyl)benzoyl)piperidine-4carboxylate (2g)

To a solution of **2f** (100 mg, 0.30 mmol, 1 eq.) in THF:MeOH 3:1 (8 mL) aqueous 1N NaOH (0.45 mL, 0.45 mmol, 1.5 eq.) was added. The mixture was stirred at room temperature overnight. The solvent was evaporated and the residue suspended in water and washed with AcOEt. The aqueous phase was acidified with HCl 1N. The precipitate formed was collected, washed with water and

diethyl ether to give 85 mg of a white solid, used in the next step without further purification. 40 mg of carboxylic acid (0.13 mmol, 1 eq.) were solubilized in DMF at 0 °C. EDC (1 eq., 0.13 mmol, 24 mg), HOBt (1 eq. 0.13 mmol, 16 mg) and methyl isonipecotate (1 eq., 0.13 mmol, 18 mg) were sequentially added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in AcOEt and washed with saturated solution of K₂CO₃, saturated solution of NH₄Cl, brine, dried over anhydrous Na₂SO₄ and concentrated. The residue was crystalized from diethyl ether to give 23 mg of a pale-yellow solid. 40 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 1.50 (m, 2H), 1.85 (m, 2H), 2.59 – 2.72 (m, 1H), 2.96 (m, 2H), 3.62 (s, 3H), 4.05 – 4.46 (m, 4H), 7.03 – 7.48 (m, 5H), 7.53 (d, *J* = 3.2 Hz, 2H), 7.68 (dd, *J* = 1.9, 17.6 Hz, 1H), 8.03 (d, *J* = 8.3 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 28.29, 36.18, 40.48, 44.44, 51.30, 120.30, 121.58, 126.61, 126.83, 128.85, 133.74, 136.79, 137.14, 139.62, 152.70, 166.30, 170.27, 175.54. m.p. [153.2 – 154.1°C]. *k*' = 7.84. HRMS m/z [M+H]⁺ Calcd. for C₂₂H₂₃N₃O₃S₂: 441.1181. Found: 441.1180.

Synthesis of 4-(((2-aminobenzo[d]thiazol-6-yl)thio)methyl)-N-benzylbenzamide (2h)

40 mg of the free carboxylic acid **2f** (0.13 mmol, 1 eq.) were solubilized in DMF at 0 °C. EDC (1 eq., 0.13 mmol, 24 mg), HOBt (1 eq. 0.13 mmol, 16 mg) and benzylamine (1 eq., 0.13 mmol, 14 mg) were sequentially added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in AcOEt and washed with K₂CO₃ saturated solution, NH₄Cl saturated solution, brine, dried over anhydrous Na₂SO₄ and concentrated. The residue was crystalized from diethyl ether to give 34 mg of a pale yellow solid. 65 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.20 (s, 2H), 4.46 (d, *J* = 5.9 Hz, 2H), 7.09 – 7.28 (m, 3H), 7.28 – 7.46 (m, 6H), 7.54 (s, 2H), 7.69 (d, *J* = 1.8 Hz, 1H), 7.74 – 7.91 (m, 2H), 8.98 (t, *J* = 6.0 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 38.51 (overlapping with DMSO peaks), 42.54, 117.85, 123.43, 125.78, 126.65, 127.15, 127.22, 128.21, 128.64, 128.78, 131.76, 132.86, 139.63, 141.48, 152.08, 165.85, 166.86. m.p. [184.4 – 185.9 °C]. *k*' = 8.78. HRMS m/z [M+H]⁺ Calcd. for C₂₂H₁₉N₃OS₂: 405.0970. Found: 405.0970.

General procedure for the synthesis of sulfone-benzothiazoles 3c-g

To a solution of benzothiazoles 2c-g (1 eq.) in DCM at 0 °C, *m*-CPBA (2.5 eq.) was added. The suspension was reacted at room temperature overnight. The reaction was diluted with DCM and washed with K₂CO₃ saturated solution. The organic phase was dried over anhydrous Na₂SO₄ and concentrated. The residue was crystalized from diethyl ether to give the desired product.

6-(benzylsulfonyl)benzo[d]thiazol-2-amine (3c)

Pale yellow solid. 87 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.60 (s, 2H), 7.14 (dd, *J* = 1.9, 7.7 Hz, 2H), 7.23 – 7.35 (m, 3H), 7.39 (d, *J* = 8.5 Hz, 1H), 7.47 (dd, *J* = 2.0, 8.5 Hz, 1H), 7.85 – 8.22 (m, 3H). ¹³C NMR (101 MHz, DMSO) δ 61.27, 116.97, 121.60, 125.82, 128.12, 128.19, 129.00, 129.71, 130.90, 131.21, 156.87, 170.25. m.p. [184.5 – 186.2°C]. *k*' = 5.65. HRMS m/z [M+H]⁺ Calcd. for C₁₄H₁₂N₂O₂S₂: 304.0340. Found: 304.0345.

6-((3,4-dichlorobenzyl)sulfonyl)benzo[d]thiazol-2-amine (3d)

Pale yellow solid. 74 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.74 (s, 2H), 7.16 (dd, *J* = 2.0, 8.3 Hz, 1H), 7.37 – 7.71 (m, 4H), 8.01 – 8.24 (m, 3H). ¹³C NMR (101 MHz, DMSO) δ 59.79, 117.05, 121.73, 125.87, 129.16, 130.27, 130.35, 130.67, 131.08, 131.13, 131.34, 132.72, 157.10, 170.43. m.p. [238°C with dec.]. k' = 6.81. HRMS m/z [M+H]⁺ Calcd for C₁₄H₁₀(³⁵Cl)₂N₂O₂S₂: 371.9561. Found: 371.9560. Calcd. for C₁₄H₁₀(³⁵Cl)(³⁷Cl)N₂O₂S₂: 373.9531. Found: 373.9530.

4-(((2-aminobenzo[d]thiazol-6-yl)sulfonyl)methyl)benzonitrile (3e)

Pale yellow solid. 79 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 4.67 (d, J = 1.5 Hz, 2H), 7.21-7.23 (m, 2H), 7.31 – 7.35 (m, 1H), 7.39-7.42 (m, 1H), 7.72-7.79 (m, 5H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 62.92, 110.45, 119.13, 119.34, 122.05, 126.55, 129.51, 130.82, 131.03, 133.48, 138.56, 157.87, 170.00. m.p. [223°C with dec.]. k' = 4.86. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₁N₃O₂S₂: 329.0293. Found: 329.0295. Pale yellow solid. 61 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.84 (s, 3H), 4.74 (s, 2H), 7.28-7.30 (m, 3H), 7.39 (d, *J* = 8.4 Hz, 1H), 7.46 (dd, *J* = 2.0, 8.5 Hz, 1H), 7.87 (d, *J* = 8.3 Hz, 2H), 8.05 (d, *J* = 1.9 Hz, 2H). ¹³C NMR (101 MHz, DMSO) δ 52.15, 60.95, 116.99, 121.63, 125.84, 128.89, 129.32, 129.46, 131.26, 131.29, 134.46, 157.01, 165.88, 170.36. m.p. [251.3 – 252.8°C]. *k*' = 5.06. HRMS m/z [M+H]⁺ Calcd. for C₁₆H₁₄N₂O₄S₂: 362.0395. Found: 362.0395.

methyl 1-(4-(((2-aminobenzo[d]thiazol-6-yl)sulfonyl)methyl)benzoyl)piperidine-4-carboxylate (3g)

White solid. 43 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 1.51 (m, 2H), 1.86 (m, 2H), 2.64 (m, 1H), 2.94 (m, 2H), 3.63 (s, 3H), 4.26 (m, 2H), 4.66 (s, 2H), 7.11 – 7.32 (m, 5H), 7.39 (d, *J* = 8.6 Hz, 1H), 7.43 – 7.55 (m, 1H), 7.93 – 8.16 (m, 3H). m.p. [232.1 – 235.1°C]. *k*' = 5.93. HRMS m/z [M+H]⁺ Calcd. for C₂₂H₂₃N₃O₅S₂: 473.1079. Found: 473.1070.

General procedure for the synthesis of amide 4c-f, i-j, l-r.

2-amino-6-carboxy-benzothiazole (1 eq.) were solubilized in DMF at 0 °C. EDC (1 eq.), HOBt (1 eq.) and the appropriate amine **9c-f**, **i-j**, **l-r** (1 eq.) were consequentially added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in AcOEt and washed consecutively with K_2CO_3 saturated solution, NH₄Cl saturated solution, brine, dried over anhydrous Na₂SO₄ and concentrated. The crude was purified on silica gel to afford the desired product.

2-amino-N-benzylbenzo[d]thiazole-6-carboxamide (4c)

White solid. 65 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.48 (d, *J* = 6.0 Hz, 2H), 7.03 – 7.48 (m, 6H), 7.66 – 7.84 (m, 3H), 8.21 (d, *J* = 1.8 Hz, 1H), 8.90 (t, *J* = 6.0 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 42.60, 116.81, 120.30, 125.04, 126.61, 126.83, 127.16, 128.19, 130.81, 139.85, 155.22,

 165.89, 168.43. m.p. [245.3 – 247.1°C]. k' = 6.17. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₃N₃OS: 283.0779. Found: 283.0780.

2-amino-N-(3,4-dichlorobenzyl)benzo[d]thiazole-6-carboxamide (4d)

White solid. 61 % yield1H NMR (400 MHz, DMSO- d_6) δ 4.44 (d, J = 1.3 Hz, 2H), 6.58 (s, 2H), 7.05 – 7.21 (m, 1H), 7.28 – 7.52 (m, 2H), 7.58 (d, J = 1.4 Hz, 2H), 8.05 (t, J = 1.0 Hz, 1H), 9.00 (s, 1H). ¹³C-NMR (101 MHz, DMSO) δ : 43.96, 118.63, 121.08, 123.23, 127.53, 129.54, 129.60 (overlapping of 2C signals), 130.30, 130.80, 131.62, 140.13, 155.96, 167.07, 169.16. m.p. [243.5 – 245.2°C]. k' = 7.16. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₁(³⁵Cl)₂N₃OS: 351.0000. Found: 351.0000. Calcd. for C₁₅H₁₁(³⁵Cl)(³⁷Cl)N₃OS: 352.9970. Found: 352.9965.

2-amino-N-(4-cyanobenzyl)benzo[d]thiazole-6-carboxamide (4e)

White solid. 45 % yield. ¹H-NMR (DMSO – *d*₆) δ: 4.50 (s, 2H), 6.66 (s, 2H), 7.53-7.55 (m, 2H), 7.67-7.77 (m, 4H), 8.37 (s, 1H), 9.08 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 42.45, 109.40, 116.85, 118.89, 120.37, 125.06, 126.47, 127.97, 130.87, 132.20, 145.79, 155.37, 166.13, 168.54. m.p. [253.6-254.7°C]. *k*' = 5.39. HRMS m/z [M+H]⁺ Calcd. for C₁₆H₁₂N₄OS: 308.0732. Found: 308.0730.

methyl 4-((2-aminobenzo[d]thiazole-6-carboxamido)methyl)benzoate (4f)

White solid. 53 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.85 (s, 3H), 4.55 (d, J = 5.9 Hz, 2H), 7.37 (d, J = 8.4 Hz, 1H), 7.42 – 7.51 (m, 2H), 7.71 – 7.86 (m, 3H), 7.86 – 8.01 (m, 2H), 8.22 (d, J =1.9 Hz, 1H), 8.99 (t, J = 6.0 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 42.44, 52.00, 116.84, 120.36, 125.07, 126.58, 127.30, 128.02, 129.18, 130.86, 145.56, 155.33, 166.03, 166.08, 168.50. m.p. [246.0 – 247.1°C]. k' = 6.00. HRMS m/z [M+H]⁺ Calcd. for C₁₇H₁₅N₃O₃S: 341.0834. Found: 341.0834.

2-amino-N-(3,4,5-trimethoxybenzyl)benzo[d]thiazole-6-carboxamide (4i)

White solid. 74 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.63 (s, 3H), 3.75 (s, 6H), 4.40 (d, *J* = 5.9 Hz, 2H), 6.66 (s, 2H), 7.35 (d, *J* = 8.4 Hz, 1H), 7.62 – 7.96 (m, 3H), 8.21 (d, *J* = 1.8 Hz, 1H), 8.87 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 42.90, 55.80, 59.93, 104.77, 116.77, 120.34, 125.10, 126.84, 130.79, 135.51, 136.29, 152.70, 155.20, 165.88, 168.43. m.p. [225°C with dec.]. *k*' = 6.69. HRMS m/z [M+H]⁺ Calcd. for C₁₈H₁₉N₃O₄S: 373.1096. Found: 373.1095.

2-amino-N-(2-nitrobenzyl)benzo[d]thiazole-6-carboxamide (4j)

Pale yellow solid. 36 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.76 (d, *J* = 5.7 Hz, 2H), 7.38 (d, *J* = 8.4 Hz, 1H), 7.48 – 7.66 (m, 2H), 7.66 – 7.95 (m, 4H), 8.05 (dd, *J* = 1.3, 8.1 Hz, 1H), 8.22 (d, *J* = 1.8 Hz, 1H), 8.98 (t, *J* = 5.8 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 40.13 (overlapping with DMSO signal), 116.86, 120.41, 124.45, 125.13, 126.29, 128.10, 129.23, 130.87, 133.68, 134.43, 148.00, 155.44, 166.29, 168.57. m.p. [240°C with dec]. *k*' = 5.60. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₂N₄O₃S: 328.0630. Found: 328.0630.

2-amino-N-benzhydrylbenzo[d]thiazole-6-carboxamide (4l)

White solid. 70 % yield. ¹H-NMR (400 MHz, DMSO- d_6) δ : 6.42 (d, J = 8.8 Hz, 2H), 7.24-7.28 (m, 2H), 7.33-7.39 (m, 8H), 7.75 (s, 2H), 7.83 (dd, J = 8.4, 1.2 Hz, 1H), 8.28 (d, J = 1.2 Hz, 1H), 9.09 (d, J = 8.8 Hz, 1H). ¹³C-NMR (101 MHz, DMSO) δ : 56.32, 116.76, 120.58, 125.60, 126.70, 126.88, 127.60, 128.26, 130.65, 142.47, 155.31, 165.54, 168.5. m.p [258°C with dec.]. k' = 8.00. HRMS m/z [M+H]⁺ Calcd. for C₂₁H₁₇N₃OS: 359.1092. Found: 359.1090.

methyl 1-(2-aminobenzo[d]thiazole-6-carbonyl)piperidine-4-carboxylate (4m)

White solid. 81 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 1.51-1.58 (m, 2H), 1.85-1.88 (m, 2H), 2.63-2.69 (m, 1H), 2.95-3.10 (m, 2H), 3.63 (s, 3H), 4.04 (from HSQC, 2H), 7.23 (dd, *J* = 1.8, 8.2 Hz, 1H), 7.33 (d, *J* = 8.2 Hz, 1H), 7.65 (s, 2H), 7.73 (dd, *J* = 0.5, 1.8 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 14.05, 27.90, 51.50, 59.71, 116.97, 120.13, 124.75, 128.20, 130.82, 153.81, 167.90,

169.19, 174.27. m.p. [195°C with dec.]. k' = 5.35. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₇N₃O₃S: 319.0991. Found: 319.0990. 2-amino-N-(2-(thiophen-2-yl)ethyl)benzo[d]thiazole-6-carboxamide (4n)

Pale brown solid. 35 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.07 (t, J = 7.2 Hz, 2H), 3.50 (td, J = 5.5, 7.2 Hz, 2H), 6.91 - 6.97 (m, 2H), 7.33 - 7.36 (m, 2H), 7.73 (d, J = 8.4 Hz, 3H), 8.15 (d, J =1.8 Hz, 1H), 8.50 (t, J = 5.6 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 29.28, 40.99, 116.79, 120.17, 123.97, 124.94, 125.07, 126.86, 127.00, 130.78, 141.61, 155.15, 165.91, 168.37. m.p. [230.5 -231.7°C]. k' = 6.27. HRMS m/z [M+H]⁺ Calcd. for C₁₄H₁₃N₃OS₂: 303.0500. Found: 303.0500.

N-(2-(1H-indol-3-yl)ethyl)-2-aminobenzo[d]thiazole-6-carboxamide (40)

White solid. 59 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 2.94 (t, J = 7.1 Hz, 2H), 3.52 (t, J = 7.0 Hz, 2H), 6.66 (s, 2H), 7.02 (dd, J = 1.6, 32.8 Hz, 2H), 7.16 (s, 1H), 7.32 (dd, J = 1.6, 7.4 Hz, 1H), 7.53 (dd, J = 1.5, 7.2 Hz, 1H), 7.66 (d, J = 1.4 Hz, 2H), 8.13 (t, J = 1.0 Hz, 1H), NH of amide exchanges, 10.79 (s, 1H). ¹³C-NMR (101 MHz, DMSO) δ 25.33, 40.33, 111.46, 112.50, 118.63, 118.77, 119.45, 121.08, 121.28, 122.60, 123.23, 127.64, 129.54, 131.62, 136.52, 155.96, 167.42, 169.16. m.p. [259°C with dec.]. k' = 6.26. HRMS m/z [M+H]⁺ Calcd. for C₁₈H₁₆N₄OS: 336.1045. Found: 336.1045.

2-amino-N-benzyl-N-(3,4-dichlorobenzyl)benzo[d]thiazole-6-carboxamide (4p)

Pale yellow solid. 47 % yield. ¹H-NMR (400 MHz, DMSO-d₆) δ: 4.71 (s, 4H), 6.66 (s, 2H), 7.08-7.10 (m, 1H), 7.23-7.37 (m, 7H), 7.52 (d, J=8.0 Hz, 1H), 7.67 (d, J=8.0 Hz, 1H), 7.99 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 51.95, 52.13, 118.92, 121.36, 122.31, 127.64, 127.76, 127.96, 128.16, 128.25, 129.60, 130.11, 130.24, 131.38, 133.25, 137.07, 137.84, 156.17, 169.16, 169.43. m.p. $[205^{\circ}C \text{ with dec.}]$. k' = 9.14. HRMS m/z $[M+H]^+$ Calcd for $C_{22}H_{17}(^{35}Cl)_2N_3OS$: 441.0469. Found: 441.0470. Calcd. for $C_{22}H_{17}({}^{35}Cl)({}^{37}Cl)N_3OS$: 443.0440. Found: 443.0440.

2-amino-N-benzyl-N-(2-hydroxyethyl)benzo[d]thiazole-6-carboxamide (4q)

White solid. 35 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 3.40-3.55 (m, 4H), 4.55-4.75 (m, 2H), 4.78 (t, J = 5.4 Hz, 1H), 7.03 – 7.47 (m, 7H), 7.64 (s, 2H), 7.80 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 30.65, 39.81 (overlapping with DMSO signals), 58.33, 116.96, 119.95, 124.60, 126.92, 127.12, 128.51, 128.86, 130.77, 137.77, 153.54, 167.74, 171.14. m.p. [178°C with dec.]. k' = 5.74. HRMS m/z [M+H]⁺ Calcd. for C₁₇H₁₇N₃O₂S: 327.1041. Found: 327.1045.

2-amino-N-(3,4-dichlorobenzyl)-N-(3,4-dimethoxybenzyl)benzo[d]thiazole-6-carboxamide (4r)

Pale yellow solid. 52 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.69 (s, 3H), 3.71 (s, 3H), 4.50 (s, 4H), 6.73 (bs, 2H), 6.90 (d, *J* = 8.2 Hz, 1H), 7.33-7.44 (m, 4H), 7.59 (d, *J* = 8.2 Hz, 1H), 7.68 (s, 2H), 7.85 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 52.20 (overlapping of 2C signals), 55.31, 55.50, 111.86, 112.45, 117.07, 120.07, 124.31, 124.43, 127.64, 128.04, 128.92, 19.57, 130.27, 130.58, 131.03, 133.41, 137.75, 148.11, 148.79, 153.90, 167.98, 171.28. m.p. [155.9-157.2°C]. *k*' = 9.42. HRMS m/z [M+H]⁺ Calcd for C₂₄H₂₁(³⁵Cl)₂N₃O₃S: 501.0681. Found: 501.0680. Calcd. for C₂₄H₂₁(³⁵Cl)(³⁷Cl)N₃O₃S: 503.0651. Found: 503.0650.

General procedure for the synthesis of benzothiazoles 4s-t

2-amino-4-hydroxy-6-carboxy-benzothiazole (1 eq.) was solubilized in DMF at 0 °C. EDC (1 eq.), HOBt (1 eq.) and the appropriate benzyl amine (1 eq.) was added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in AcOEt and washed consecutively with K_2CO_3 saturated solution, NH₄Cl saturated solution, brine, dried over anhydrous Na₂SO₄ and concentrated. The crude was purified on silica gel to afford the desired product.

2-amino-N-benzyl-4-hydroxybenzo[d]thiazole-6-carboxamide (4s)

Pink solid. 23 % yield. ¹H-NMR (400 MHz, DMSO-*d*₆)) δ: 4.44 (d, 2H, *J* = 6 Hz), 7.24 (m, 2H), 7.32 (m, 4H), 7.51 (s, 2H), 7.68 (s, 1H), 8.80 (t, 1H, *J* = 6 Hz), 9.44 (s, 1H). ¹³C-NMR (400 MHz, DMSO-*d*₆) δ: 42.57, 111.00, 11.18, 126.60, 127.15, 127.97, 128.18, 131.55, 139.90, 143.85, 146.92,

 166.08, 166.29. m.p. [244°C with dec.]. k' = 6.39. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₃N₃O₂S: 299.0728. Found: 299.0725.

2-amino-N-(3,4-dichlorobenzyl)-4-hydroxybenzo[d]thiazole-6-carboxamide (4t)

Yellow solid. 32 % yield. ¹H-NMR (400 MHz, DMSO-*d*₆) δ : 4.43 (d, 2H, *J* = 5.6 Hz), 7.21 (s, 1H), 7.30 (dd, 1H, *J*=8.4, 2.0 Hz), 7.53 (s, 2H), 7.55 (d, 1H, *J*=1.6 Hz), 7.59 (d, 1H, *J* = 8.4 Hz), 7.67 (d, 1H, *J* = 1.6 Hz), 8.86 (t, 1H, *J* = 6 Hz), 9.45 (s, 1H). ¹³C-NMR (400 MHz, DMSO-*d*₆) δ : 41.66, 111.12, 117.74, 125.57, 127.60, 129.19, 2x130.42, 130.79, 131.61, 141.22, 144.08, 146.95, 166.21, 166.37. m.p. [251°C with dec.]. k' = 5.41. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₁(³⁵Cl)₂N₃O₂S: 366.9949. Found: 366.9950. Calcd for C₁₅H₁₁(³⁵Cl)(³⁷Cl)N₃O₂S: 368.9920. Found: 368.9920.

General procedure for the synthesis of the 4-amido-benzothiazoles 5a-c

To a solution of **13a-c** (1 eq.) in dry DCM (20 mL) at 0 °C under an inert atmosphere a solution 1M of BBr₃ (3 eq. *per* CH₃O) in DCM was added drop wise. The reaction was spontaneously warmed at room temperature and stirred for 4 hours. The mixture was quenched with 10 mL of MeOH and concentrated. The residue was suspended in water and extracted in AcOEt. The organic phase was washed with NaHCO₃ saturated solution and brine, dried over anhydrous Na₂SO₄ and concentrated. The crude was purified over silica gel to give the desired product.

2-amino-N-benzyl-N-(3,4-dichlorobenzyl)-6,7-dihydroxybenzo[d]thiazole-4-carboxamide (5a)

Purple solid. 68 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.33 (s, 2H), 4.59 (s, 2H), 6.71 (s, 2H), 6.96 – 7.45 (m, 6H), 7.46 – 7.71 (m, 2H), 7.74 (d, *J* = 8.3 Hz, 1H), 9.15 (bs, 1H), 9.43 (bs, 1H). ¹³C NMR (101 MHz, DMSO) δ 45.82, 51.84, 112.8, 116.45, 127.33, 127.43, 127.51, 128.47, 129.12, 129.22 (overlap of 2C signals), 130.43, 130.72, 130.86, 136.90, 138.74, 139.03, 139.07, 165.24, 169.86. m.p. [215°C with dec.]. k' = 7.80. HRMS m/z [M+H]⁺ Calcd. for C₂₂H₁₇(³⁵Cl)₂N₃O₃S: 473.0368. Found: 473.0365. Calcd. for C₂₂H₁₇(³⁵Cl)(³⁷Cl)N₃O₃S: 475.0338. Found: 475.0340.

2-amino-N-(3,4-dichlorobenzyl)-6,7-dihydroxybenzo[d]thiazole-4-carboxamide (5b)

Green solid. 62 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.57 (d, J = 6.1 Hz, 2H), 7.30 (dd, J = 2.1, 8.3 Hz, 1H), 7.50-7.74 (m, 6H), 9.36 (s, 1H), 10.19 (s, 1H). ¹³C NMR (101 MHz, DMSO) δ 41.24, 109.16, 115.25, 118.90, 124.74, 126.76, 126.81, 127.51, 128.02, 137.34, 143.53, 144.85, 145.23, 161.7, 164.07. m.p. [170°C with dec.]. k' = 5.69. HRMS m/z [M+H]⁺ Calcd for C₁₅H₁₁(³⁵Cl)₂N₃O₃S: 382.9898. Found: 382.9900. Calcd. for C₁₅H₁₁(³⁵Cl)(³⁷Cl)N₃O₃S: 384.9869. Found: 384.9870.

2-amino-N-benzyl-6,7-dihydroxybenzo[d]thiazole-4-carboxamide (5c)

Dark yellow solid. 70 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.60 (d, J = 6.1 Hz, 2H), 7.15 – 7.45 (m, 5H), 7.52 (s, 1H), 7.71 (s, 2H), 9.32 (s, 1H), 9.82 (s, 1H), 10.17 (t, J = 6.2 Hz, 1H). ¹³C NMR (101 MHz, DMSO) δ 42.11, 112.42, 114.82, 117.89, 126.64, 126.83, 128.36, 138.96, 139.09, 139.88, 142.41, 164.55, 166.38. m.p. [192°C with dec.]. k' = 4.73. HRMS m/z [M+H]⁺ Calcd. for C₁₅H₁₃N₃O₃S: 315.0678. Found: 315.0678.

6-hydroxybenzo[d]thiazol-2-amine (6)

To a solution of thiourea (500 mg, 6.58 mmol, 1 eq.) in ethanol (10 mL), concentrated hydrochloric acid (55 μ L, 0.66 mmol, 0.1 eq.) was added, followed by the drop wise addition of a hot solution of benzoquinone (1448 mg, 13.66 mmol, 2 eq.) in ethanol over a period of 30 minutes. The solution was refluxed for 6 hours and then concentrated. The crude was titrated with hot acetonitrile and filtered. The solid was re-suspended in water and neutralized with sodium acetate. A purple solid precipitated, which was collected, washed with hot acetone and filtered to give 830 mg of a brick-red solid (76 % yield). ¹H-NMR (400 MHz, DMSO- *d*₆) δ 6.63 (dd, *J* = 8.4, 2.4 Hz, 1H), 7.00 (d, *J* = 2.4, 1H), 7.06 (s, 2H), 7.11(d, *J* = 8.4, 1H), 9.060 (s, 1H). ¹³C-NMR (101 MHz, DMSO) δ 107.36, 113.97, 118.55, 132.31, 146.13, 152.54, 164.40.

General procedure for the synthesis of nitrobenzene-sulfides 7a-f

To a solution of 4-nitro-thiophenol (1 eq.) in DMF (5 mL per mmol) K_2CO_3 (2.5 eq.) was added at room temperature. After 10 minutes, the appropriate alkyl bromide (1.2 eq., for **7b-h**) or iodomethane (1.2 eq., for **7a**) was added and the mixture stirred in the same conditions for 24 hours and then concentrated. The residue was suspended in AcOEt and washed with a saturated solution of K_2CO_3 , brine, dried over anhydrous Na₂SO₄ and concentrated to afford the desired product.

methyl(4-nitrophenyl)sulfane (7a)

Yellow liquid. 92 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 2.45 (s, 3H), 7.49 (d, *J* = 7.6 Hz, 2H), 8.15 (d, *J* = 7.5 Hz, 2H). MS m/z [M+H]⁺: 170.2.

(4-nitrophenyl)(trifluoromethyl)sulfane (7b)

Pale yellow solid. 84 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.57 (d, *J* = 7.4 Hz, 2H), 8.16 (d, *J* = 7.6 Hz, 2H). MS m/z [M+H]⁺: 224.3.

benzyl(4-nitrophenyl)sulfane (7c)

Yellow solid. 95 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 4.33 (s, 2H), 7.09 – 7.36 (m, 5H), 7.37 – 7.56 (m, 2H), 8.04 – 8.21 (m, 2H). MS m/z [M+H]⁺: 246.4.

(3,4-dichlorobenzyl)(4-nitrophenyl)sulfane (7d)

Yellow solid. 86 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 4.33 (s, 2H), 6.67 – 7.06 (m, 1H), 7.14 – 7.44 (m, 2H), 7.44 – 7.77 (m, 2H), 8.06 – 8.29 (m, 2H). MS m/z [M+H]⁺: 314.1, 316.2.

4-(((4-nitrophenyl)thio)methyl)benzonitrile (7e)

Yellow solid. 82 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 4.33 (s, 2H), 7.27 – 7.42 (m, 2H), 7.50-7.54 (m, 4H), 8.15 (d, *J* = 7.6 Hz, 2H). MS m/z [M+H]⁺: 271.4.

methyl 4-(((4-nitrophenyl)thio)methyl)benzoate (7f)

Yellow solid. 75 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.95 (s, 3H), 4.33 (s, 2H), 6.97 – 7.32 (m, 2H), 7.48 (d, *J* = 7.5 Hz, 2H), 7.82 (d, *J* = 7.7 Hz, 2H), 8.14 (d, *J* = 7.6 Hz, 2H). MS m/z [M+H]⁺: 304.4.

General procedure for the synthesis of aminobenzene-sulfides 8a-f

To a solution of the appropriate 4-nitrophenylsulfane **7a-f** (1 eq.) in methanol (10 mL *per* mmol) NH_4Cl (2.5 eq.) and zinc (2 eq.) were added at room temperature. The suspension was refluxed for 2-6 hours and concentrated. The residue was suspended in water and AcOEt and the aqueous phase alkalinized with 6N NaOH. The suspension was filtered over celite. The organic phase was collected, dried over anhydrous Na_2SO_4 and concentrated. The crude was purified on silica gel to afford the desired product.

4-(methylthio)aniline (8a)

Yellow liquid. 64 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 2.45 (s, 3H), 6.53 (d, *J* = 7.6 Hz, 2H), 7.07 (d, *J* = 7.6 Hz, 2H). MS m/z [M+H]⁺: 140.2.

4-((trifluoromethyl)thio)aniline (8b)

Yellow liquid. 32 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 6.54 (d, *J* = 7.6 Hz, 2H), 7.14 (d, *J* = 7.5 Hz, 2H). MS m/z [M+H]⁺: 194.1.

4-(benzylthio)aniline (8c)

Yellow solid. 75 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.33 (d, J = 1.1 Hz, 2H), 6.55 (d, J = 7.4 Hz, 2H), 6.91 – 7.36 (m, 7H), 7.34 – 7.52 (m, 2H). MS m/z [M+H]⁺: 216.2.

4-((3,4-dichlorobenzyl)thio)aniline (8d)

Yellow solid. 69 % yield. ¹H NMR (400 MHz, DMSO- d_6) δ 4.30 (m, 2H), 6.57 (d, J = 7.6 Hz, 2H), 6.92 – 7.36 (m, 5H), 7.37 – 7.56 (m, 1H), 7.62 (d, J = 7.5 Hz, 1H). MS m/z [M+H]⁺: 284.1, 286.1.

4-(((4-aminophenyl)thio)methyl)benzonitrile (8e)

Yellow solid. 58 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.35 (s, 2H), 6.54 (d, *J* = 7.4 Hz, 2H), 7.04 – 7.47 (m, 6H), 7.64 (d, *J* = 7.4 Hz, 2H). MS m/z [M+H]⁺: 241.2.

methyl 4-(((4-aminophenyl)thio)methyl)benzoate (8f)

Yellow solid. 63 % yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.85 (s, 3H), 4.32 (d, *J* = 1.1 Hz, 2H), 6.55 (d, *J* = 7.6 Hz, 2H), 6.97 – 7.42 (m, 6H), 7.82 (d, *J* = 7.6 Hz, 2H). MS m/z [M+H]⁺: 274.4.

General procedure for the synthesis of amine 9p-r

To a solution of the amine (2 eq.) in DMF, K_2CO_3 (2.5 eq.) and the appropriate alkyl bromide were added at room temperature. The mixture was stirred in the same condition for 6 hours and concentrated. The residue was suspended in DCM and washed with brine. The organic phase was dried over anhydrous Na_2SO_4 and concentrated. The crude was chromatographed to afford the desired product.

N-benzyl-1-(3,4-dichlorophenyl)methanamine (9p)

Yellowish liquid. 35 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.65 (s, 4H), 7.18 – 7.39 (m, 6H), 7.39 – 7.49 (m, 2H). MS m/z [M+H]⁺: 265.1; 267.1.

2-(benzylamino)ethan-1-ol (9q)

Colorless liquid. 58 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 2.74 (t, *J* = 4.5 Hz, 2H), 3.69 (q, *J* = 4.7 Hz, 2H), 4.25 (d, *J* = 1.1 Hz, 2H), 7.19 - 7.41 (m, 3H), 7.47 - 7.61 (m, 2H). MS m/z [M+H]⁺: 152.2.

N-(3,4-dichlorobenzyl)-1-(3,4-dimethoxyphenyl)methanamine (9r)

Yellowish liquid. 48 % yield. 1H NMR (400 MHz, Chloroform-d) δ 3.64 (s, 2H), 3.66 (s, 2H), 3.88 (s, 3H), 3.90 (s, 3H), 6.84 (d, *J* = 7.4 Hz, 1H), 6.90 (dd, *J* = 1.6, 7.5 Hz, 1H), 7.06 (dt, *J* = 1.1, 2.1 Hz, 1H), 7.22 – 7.29 (m, 1H), 7.37 – 7.47 (m, 2H). MS m/z [M+H]⁺: 326.1; 327.2.

Synthesis of 2-amino-4-hydroxybenzo[d]thiazole-6-carboxylic acid (10)

To a suspension of 2-amino-4-hydroxybenzoic acid (500 mg, 3.26 mmol, 1 eq.) in glacial acetic acid (6 mL), KSCN (1.6 g, 16.32 mmol, 5 eq.) was added. The mixture was chilled and a solution of elemental bromine (336 μ L, 6.52 mmol, 2 eq.) in acetic acid (6 mL) was added drop wise keeping the temperature below 10 °C. The mixture was spontaneously warmed at room temperature and stirred for 1 hour. The reaction was quenched with water, boiled for 15 minutes and filtered still hot. The filtrate was chilled down and the solid crystalized removed by filtration. The pH of the water was adjusted to 4 and the solid precipitated collected by filtration, rinsed with water and dried to afford 290 mg (42 % yield) of a dark orange solid. ¹H NMR (400 MHz, DMSO-*d*₆) δ 6.74 (s, 2H), 7.63 (d, *J* = 1.7 Hz, 1H), 8.01 (d, *J* = 1.5 Hz, 1H), 10.25 (s, 1H), 12.99 (s, 1H). MS m/z [M+H]⁺: 211.0.

Synthesis of methyl 2-amino-6,7-dimethoxybenzo[d]thiazole-4-carboxylate (11)

To a suspension of methyl 2-amino-4,5-dimethoxybenzoate (1.0 g, 4.73 mmol, 1 eq.) in glacial acetic acid (20 mL), KSCN (2.3 g, 23.7 mmol, 5 eq.) was added. The mixture was chilled and a solution of elemental bromine (488 μ L, 9.46 mmol, 2 eq.) in acetic acid (10 mL) was added drop wise keeping the temperature below 10 °C. The mixture was spontaneously warmed at room temperature and stirred for 2 hours. The mixture was quenched with water and filtered. The filtrate was alkalinized at pH 9 with 30% aqueous NH₄OH and the precipitated formed collected by filtration, rinsed with water and dried to give 1.02 g (yield 81 %) of a brown solid. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.81 (s, 3H), 3.84 (s, 3H), 3.92 (s, 3H), 7.39 (s, 1H), 7.76 (s, 2H). MS m/z [M+H]⁺: 269.1.

2-amino-6,7-dimethoxybenzo[d]thiazole-4-carboxylic acid (12)

To a solution of **11** (500 mg, 1.86 mmol, 1 eq.) in MeOH:THF 3:1 (12 mL) aqueous 1N NaOH (2.05 mL, 2.05 mmol, 1.1 eq.) was added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in water and washed with AcOEt. The pH of the aqueous phase was adjusted to 4 with 1N HCl and the precipitate formed was collected by filtration, rinsed with water and dried to give 276 mg (58 % yield) of a brown solid. ¹H NMR (400 MHz, DMSO-*d*₆) δ 3.81 (s, 3H), 3.85 (s, 3H), 7.56 (s, 1H), 8.60 (bs, 2H). MS m/z [M-H]⁻: 253.0.

General procedure for the synthesis of the 4-amido-benzothiazoles 13a-c

12 (1 eq.) was solubilized in DMF at 0 °C. EDC (1 eq.), HOBt (1 eq.) and the appropriate benzyl amine (1 eq.) were consequentially added. The mixture was stirred at room temperature overnight and concentrated. The residue was solubilized in AcOEt and washed consecutively with K_2CO_3 ss, NH₄Cl ss, brine, dried over anhydrous Na₂SO₄ and concentrated. The crude was purified on silica gel to afford the desired product.

2-amino-N-benzyl-N-(3,4-dichlorobenzyl)-6,7-dimethoxybenzo[d]thiazole-4-carboxamide

(13a)

Brown solid. 61 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.90 (s, 3H), 3.96 (s, 3H), 4.70-4.72 (m, 4H), 6.97 – 7.11 (m, 1H), 7.18 – 7.37 (m, 6H), 7.44 (d, *J* = 7.5 Hz, 1H), 7.66 (s, 1H). MS m/z [M+H]⁺: 502.1.

2-amino-N-(3,4-dichlorobenzyl)-6,7-dimethoxybenzo[d]thiazole-4-carboxamide (13b)

Brown solid. 39 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.90 (s, 3H), 3.91 (s, 3H), 4.52 (d, *J* = 1.3 Hz, 2H), 7.33 (dd, *J* = 1.5, 7.2 Hz, 1H), 7.46 – 7.62 (m, 2H), 8.11 (s, 1H). MS m/z [M+H]⁺: 413.1.

2-amino-N-benzyl-6,7-dimethoxybenzo[d]thiazole-4-carboxamide (13d)

Brown solid. 48 % yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.89 (s, 3H), 3.90 (s, 3H), 4.50 (s, 2H), 7.12 – 7.44 (m, 5H), 7.65 (s, 1H). MS m/z [M+H]⁺: 345.0.

Molecular modelling

The crystal structure of *Tb*PTR1 in ternary complex with NAPD(H) and DX2 inhibitor at 1.8 Å was retrieved from Protein Data Bank (PDB ID: 3JQ7).²⁸ Chain A was prepared for calculations by using VMD.³⁹ Missing bonds and atoms were fixed and polar hydrogens and charges were added. Conserved water molecules previously identified for TbPTR1 and a set of additional waters present in at least 50 % of the *Tb*PTR1 crystal structures reported in literature were included in the binding site.¹⁷ A three-dimensional grid box was centered on the oxygen atom of hydroxyl group of Tyr174. The center and the size of the grid maps were based on AutoGrid default values (Size (x,y,z): 40/50/40 grid points spacing of 0.375 Å; Center (x,y,z): 6.798/-6.705/16.560). AutoDock 4.2 was used for the docking simulation.²⁹ Lamarckian genetic algorithm (LGA) was employed for ligand conformational searching and docking parameters were set as default. The docking model was validated first by redocking of the ligand DX2 into the prepared crystal structure with the resulting RMSD being within 0.83 Å. Ligands were first drawn with ChemBio3D Ultra 14.0,40 and all the combinations of tautomers and protonation states at pH 7.4 of the ligands were generated with Discovery Studio.⁴¹ AutoDockTools 1.5.6²⁹ package scripts were used to convert structures of ligands and prepared protein to AutoDock 4.2 format. A total of 20 conformations with the best scores were retained for each ligand. Finally, docking poses were ranked according to the predicted scores and visually inspected with AutodockTools 1.5.629 and Discovery Studio software.41

Protein expression and purification

Recombinant TbPTR1 was expressed and purified by established methods.²⁴

Protein crystallization

Crystals of *Tb*PTR1 were grown using the vapor diffusion hanging drop technique at 297 K.⁴² Drops were prepared by mixing equal volumes (3 μ L) of protein (histidine tagged *Tb*PTR1 8-10 mg mL⁻¹ in 20 mM TRIS, pH 7.5) and precipitant (2 – 2.5 M sodium acetate, 0.1 M sodium citrate, pH 5) solutions. The ternary complexes *Tb*PTR1-cofactor-inhibitor were obtained by the soaking technique using preformed monoclinic protein crystals, grown after few days of equilibration at 297 K. The complexes with the compounds **3a**, **4c** and **4g** were obtained by transferring crystals to a cryo-protectant/soaking solution prepared by adding 30 % v/v glycerol and 4 mM inhibitor (solubilized in DMSO) to the precipitant solution (the final DMSO concentration was kept lower than 10 % v/v), according to an established procedure.²² Crystals were flash-frozen in liquid nitrogen after 7 – 20 hours. All the other complexes were obtained by adding 2 – 4 mM inhibitor (solubilized in DMSO) in the crystallization drop (the DMSO concentration in the drop was kept lower than 10 % v/v), as previously reported.¹⁷ After 1 – 2 hours crystals were transferred to the cryo-protectant solution (obtained by adding 30 % v/v glycerol to the precipitant solution) and flash-frozen in liquid nitrogen.

Data collection, structure solution and refinement

Diffraction data were collected using the synchrotron radiation either at the Diamond Light Source (DLS, Didcot, United Kingdom) beam line I04 and I04-1 equipped with a Dectris Pilatus 6M-F detector or at the European Synchrotron Radiation Facility (ESRF, Grenoble, France) beam line ID23-1 and ID23-2 equipped with a Pilatus 6M-F and a Pilatus3 X 2M, respectively. Reflections were integrated using either Mosflm ^{43,44} or XDS ⁴⁵ and scaled with Scala ^{46,47} from the CCP4 suite.⁴⁸ The *Tb*PTR1 crystals belonged to the primitive monoclinic space group P2₁, with a functional tetramer in the asymmetric unit. Data collection and processing statistics are reported in Table SI-2-3. Structures were solved by the molecular replacement technique using the software MOLREP ⁴⁹ and a functional tetramer of *Tb*PTR1 (PDB ID 5JDC ²⁴) as the searching model (solvent and non-protein molecules were preventively excluded). The protein structures were

refined using REFMAC5⁵⁰ and a protocol involving sequential iterative manual rebuilding of the models and maximum likelihood refinement. The molecular graphic software Coot ^{51,52} was used for visual inspection and modeling of missing atoms in the electron density. Ligand occupancy was estimated using a criterion based on keeping the atomic displacement parameters of each molecule comparable with those of the surrounding residues (in fully occupied sites). Water molecules were added using the program ARP/wARP ⁵³ and inspected with Coot. The final models were checked both manually and through the programs Coot and Procheck ⁵⁴ and rendered using CCP4mg ⁵⁵. Final refinement statistics are reported in Table SI-2.

Coordinates and structure factors were deposited in the Protein Data Bank under the codes **6GCK** (*Tb*PTR1 – NADP(H)– **1e**), **6GCQ** (*Tb*PTR1 – NADP(H)– **2b**), **6GCP** (*Tb*PTR1 – NADP(H)– **2d**), **6GDO** (*Tb*PTR1 – NADP(H)– **2g**), **6GEX** (*Tb*PTR1 – NADP(H)– **2h**), **6GCL** (*Tb*PTR1 – NADP(H)– **3a**), **6GD4** (*Tb*PTR1 – NADP(H)– **4c**), **6GEY** (*Tb*PTR1 – NADP(H)– **4d**), **6GD0** (TbPTR1 – NADP(H)– **4g**), and **6GDP** (*Tb*PTR1 – NADP(H)– **4l**).

In-vitro biological assays

*Tb***PTR1**, *Lm***PTR1** inhibition assays. The *in-vitro* assays for the detection of PTR1 inhibition, used in the current study, were based upon those reported in the literature.⁵⁶ Since PTR1 enzymes use dihydrobiopterine (H₂B) as a substrate and also require NADPH for the reaction, the reduction of H₂B to tetrahydrobiopterine (H₄B) by PTR1 is non-enzymatically linked with the reduction of cytochrome c in this assay, which is detected at 550 nm. The formation of cyt c Fe²⁺ results in a signal increase in the photometric readout.

Evaluation of activity against *T. brucei.* The anti-parasitic activity of compounds was evaluated using a modified resazurin-based assay previously described in the literature.⁵⁷ Mid-log *T. brucei* bloodstream forms were added to an equal volume of serial dilutions of compounds in

supplemented complete HMI-9 medium at a final cell density of 5 x 10^3 /mL. Following incubation for 72 hours at 37 °C 5 % CO₂, 20 µL of a 0.5 mM resazurin solution was added and plates were incubated for a further 4 hours under the same conditions. Fluorescence was measured at 540 nm and 620 nm excitation and emission wavelength, respectively, using a Synergy 2 Multi-Mode Reader (Biotek). The anti-trypanosomatid effect was evaluated by the determination of the EC₅₀ value (concentration required to inhibit growth by 50 %) and calculated by non-linear regression analysis using GraphPad Prism, Version 5.00 for Windows, GraphPad Software, San Diego California USA (www.graphpad.com). For potentiation experiments the synergy index was calculated as the ratio between the activity of the combination (10 µM of compound in the presence of 4 µM MTX) and the added activity of the components of the combination alone (activity of 10 µM of benzothiazole + activity of 4 µM MTX).

Evaluation of activity against *L. infantum* intramacrophage amastigotes. Anti-parasitic activity against *L. infantum* intracellular amastigotes at 10 μ M compound concentration was determined according to a previously published procedure with some modifications.⁵⁸ Briefly, 1 × 106 THP-1-derived macrophages were infected with luciferase-expressing *L. infantum* axenic amastigotes in a macrophage:amastigotes ratio of 1:10 for 4 h at 37 °C, 5% CO₂. Non-internalized parasites were washed, and compounds were added at different concentrations (20 - 0.6 μ M). After 72 h of incubation at 37 °C and 5% CO₂, media was substituted by PBS. Macrophages were lysed by addition of 25 μ L of Glo-lysis buffer (Promega). Steady-Glo reagent (Promega) was then added, and the content of each well was transferred to white-bottom 96-well plates. Luminescence intensity was read using a Synergy 2 multi-mode reader (Biotek). The anti-leishmanial effect was evaluated by the determination of the IC₅₀ value (concentration required to inhibit growth in 50%) and calculated by nonlinear regression analysis using GraphPad Prism version 5.00 for Windows, GraphPad Software, San Diego, California (www.graphpad.com).

Evaluation of activity against *T. cruzi.* The drug assay method consisted of measuring compound activity on the U2OS (osteosarcoma) cell-line infected with *T. cruzi*, as previously described.³⁴ The U2OS cell-line, which presents a large cytoplasm that allow for improved quantification of *T. cruzi* amastigotes in high content analysis, was plated and allowed to grow as a monolayer for 24 h prior to infection with tissue-derived trypomastigote forms of *T. cruzi*. Compounds were plated at a final concentration of either 50 μ M (for single concentration screening) or variable concentrations (for dose-response tests) 24 h after infection. Infected cultures were exposed to compounds for 72 hours. Plates were processed for high content imaging and normalized compound activity was determined in relation to infected and non-infected controls.

In-vitro early ADME-Toxicity assays

*h*ERG inhibition. This assay made use of Invitrogen's PredictorTM *h*ERG Fluorescence Polarisation Assay. The assay uses a membrane fraction containing *h*ERG channel (PredictorTM *h*ERG Membrane) and a high-affinity red fluorescent *h*ERG channel ligand, or "tracer" (PredictorTM *h*ERG Tracer Red), whose displacement by test compounds can be determined in a homogenous, Fluorescence Polarization based format.⁵⁹

Cytochrome P450 1A2, 2C9, 2C19, 2D6 and 3A4 inhibition. These assays made use of the Promega P450-GloTM assay platform. Each cytochrome P450 isoform assay made use of microsomal preparations of cytochromes from baculovirus infected insect cells. Action of the cytochrome P450 enzymes upon each substrate ultimately resulted in the generation of light and a decrease in this was indicative of inhibition of the enzymes.⁵⁹

Cytotoxicity assay against A549 cell-line. The assays were performed using the Cell Titer-Glo[®] assay from Promega. The assay detects cellular ATP content with the amount of ATP being directly proportional to the number of cells present. The A549 cell-line was obtained from DSMZ (German

Collection of Microorganisms and Cell Cultures, Braunschweig, Germany) and was grown in DMEM with FCS (10 % v/v), streptomycin (100 μ g/ml) and penicillin G (100 U/ml).⁵⁹

Cytotoxicity assessment against THP-1 macrophages and U2OS cells. The cytotoxicity of THP-1-derived macrophages was assessed by the colorimetric (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) assay.⁵⁹ The cytotoxicity against U2OS cells was determined from the anti-*T. cruzi* assay, based on the reduction of cell number in treated wells in comparison with infected, non-treated wells.³⁴

Mitochondrial toxicity. This assay made use of MitoTracker[®] Red chloromethyl-X-rosamine (CMXRos) uptake and High Content Imaging to monitor compound mediated mitochondrial toxicity in the 786-O (renal carcinoma) cell line. Cells were maintained using RPMI-1640 medium containing 2 mM glutamine, FCS (10 % v/v), streptomycin (100 μ g/ml) and penicillin G (100 U/ml).⁵⁹

Pharmacodynamic solubility. The pharmacodynamics solubility of **4c** in four media (Milli-Q water, PBS, PBS + 10%DMSO and PBS + 50%DMSO) was spectrophotometrically measured by the methodology described by Bard et al. with slight modification.³⁵ The molar extinction coefficient (ε) at 340 nm was determined first. Four over-saturated solutions, with maximum theoretical concentration of 100 µg/mL were prepared and the suspensions stirred at 1200 rpm for 24 hours at 25 °C. The suspensions were filtered (0.2 µm) and diluted 1:1 with ACN and the concentration of the compound in the filtrate was quantified spectrophotometrically.

In-vivo assays

Pharmacokinetics of compound 4c by LC-MS. The snapshot PK studies were carried out according to literature.⁶⁰ BALB/c mice were treated with compound 4c administered alone (8% DMSO) (1 mg/kg IV or 20 mg/kg per os), compound 4c solubilized with hydroxypropil-βcyclodextrin (50%) (20 mg/kg per os) or with a mixture of cyclodextrins (25%) plus polyethylene glycol 400 (Quimidroga) (50%) (20 mg/kg per os). Plasma samples were analyzed by LC-MS. Chromatographic separation was carried out using a Shimadzu LC system consisting of two pumps, column oven, degasser and autosampler. Attached to this system was an analytical column (C18, Gemini 5 µm 110 A Phenomenex, 150 x 2 mm). The HPLC system was connected to a triplequadrupole mass spectrometer equipped with a turboionspray source operated with unit resolution in the positive ion mode (ESI-QQQMS, Shimadzu LCMS-30). Under these conditions a retention time of approximately 4 minutes for the molecule was obtained. Mobile phase consisted of a mixture (80:20) of acetonitrile: purified water at a flow rate of 0.2 mL/min rate in isocratic mode. Run time was 10 minutes and the injection volume was 5 μ L. The mass transition of m/z was: Quantifier (m/z): 315.0 > 161.0 CE: - 20; Qualifier (m/z): 315.0 > 132.9 CE: -34. Plasma samples of mice (NMRI and BALB/c) were obtained by serial sampling from submandibular vein and stored at -20 °C until being analyzed.

In-vivo evaluation of anti-*T.brucei* activity of 4c. Five-week-old BALCB/c female mice were purchased from Charles River Laboratories (Saint-Germain-Nuelles, France), and housed in the animal facility of the Instituto de Investigação e Inovação em Saúde (i3S). The animals were kept in individually ventilated cages with high efficiency particulate air (HEPA) filters and were fed ad libitum with sterilized food and water for one week prior to experimental infections. The *Trypanosoma brucei brucei*, GVR35 line, expressing the red-shifted luciferase⁶¹ was kindly provided by Prof. Jeremy Mottram, York University. Mice were infected by intraperitoneal injection with blood containing 1 x 10⁵ parasites harvested at the first peak of parasitemia from a donor mouse. Control animals were similarly injected with the same volume of PBS. Compound 4c

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was administered orally as a suspension in a 2 g/mL (2-Hydroxypropyl)-β-cyclodextrin solution at the dose of 40 mg/kg twice a day in order to possibly circumvent their limited bioavailability. Pentamidine, at the dose of 5 mg/kg/day via IP administration, was used as positive control. Parasite loads were assessed by bioluminescence imaging using the IVIS Lumina LT in vivo imaging system (PerkinElmer, Waltham MA, USA). T. b. brucei GVR35-infected mice were anesthetized with 2.5% isoflurane (O2 flow of 1 L/min) and D-luciferin (2.4 mg, Perkin Elmer) was administered subcutaneously five minutes prior to image acquisition. Mice were then transferred to the stage of an intensified charge-coupled device photon-counting video camera box where anaesthesia was maintained with 2.5% isoflurane (O2 flow of 0.3 L/min). Exposure to isoflurane was standardized among groups and throughout all time points and signal acquisition was controlled by the Living Image software (Perkin Elmer). The detection of the bioluminescence signal by the system resulted in the generation of signal maps automatically superimposed to the grey-scale photograph of the mice. The regions of interest (ROI) encompassing most of the ventral view of the animal body, the head, the thorax, the liver, the spleen, the abdomen, and the lower abdomen were manually defined. The quantifications were performed using the Living Image software The total flux (photons/second) (Perkin Elmer). and average radiance (photons/second/cm²/steradian) within these ROIs were automatically calculated. The percentage of bioluminescence signal in the defined regions was calculated by dividing the total flux of the respective ROI by the total flux of the ventral animal body ROI.

ANCILLARY INFORMATION

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website, including (*i*) active site view of *Tb*PTR1 in complex with the cofactor NADP(H) and **4d**; (*ii*) calculated rule of 5 (RO5) properties; (*iii*) data collection and processing statistics; (*iv*) structure solution and 64

refinement; (*v*) correlation between the pIC50 against TbPTR1 and the Autodock Score; (*vi*) TIP criteria; (*vii*) ¹H and ¹³C-NMR spectra of the tested compounds (PDF); (*viii*) PDB files of the following docked complexes: 3jq7_1c_dockedcomplex (PDB) 3jq7_2c_dockedcomplex (PDB) 3jq7_2g_dockedcomplex (PDB) 3jq7_3c_dockedcomplex (PDB) 3jq7_4c_dockedcomplex (PDB) 3jq7_5b_dockedcomplex (PDB); and (*ix*) Molecular Formula String.

PDB ID Codes

Crystal structures of the ternary complexes of five compounds have been deposited with the following PDB accession codes: 6GCK (TbPTR1 - NADP(H) - 1e), 6GCQ (TbPTR1 - NADP(H) - 2b), 6GCP (TbPTR1 - NADP(H) - 2d), 6GDO (TbPTR1 - NADP(H) - 2g), 6GEX (TbPTR1 - NADP(H) - 2h), 6GCL (TbPTR1 - NADP(H) - 3a), 6GD4 (TbPTR1 - NADP(H) - 4c), 6GEY (TbPTR1 - NADP(H) - 4d), 6GD0 (TbPTR1 - NADP(H) - 4g), and 6GDP (TbPTR1 - NADP(H) - 4d). Authors will release the atomic coordinates and experimental data upon article publication.

Corresponding Authors

*E-mail: p.linciano@unimore.it Phone: 0039-059-205-8661 (P.L.).

*E-mail: stefano.mangani@unisi.it Phone: 0039-0577234255 (S.M.)

*E-mail: luca.costantino@unimore.it Phone: 0039-0592058572 (L.C.)

*E-mail: mariapaola.costi@unimore.it, costimp@unimore.it. Phone: 0039-059-205-8579 (M.P.C.).

ORCID

Pasquale Linciano: 0000-0003-0382-7479

Maria Paola Costi: 0000-0002-0443-5402

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Ethics Statement. All experiments with mice were carried out in accordance with the IBMC.INEB Animal Ethics Committees and the Portuguese National Authorities for Animal Health guidelines according to the statements on the directive 2010/63/EU of the European Parliament and Council.

Abbreviations

Acetonitrile: ACN; Cytochrome P450, CYP; Dihydrobiopterin: H₂B; Dihydrofolate reductase: DHFR; Ethyl acetate: AcOEt; 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide: EDC; Formic acid: FA; Human African trypanosomiasis: HAT; Hydroxybenzotriazole: HOBt; hydroxypropyl- β cyclodextrin, HP- β -CD; *Leishmania infantum, L. infantum; Leishmania major, L. major; Leishmania major* pteridine reductase 1: *Lm*PTR1; Methanol: MeOH; Methotrexate: MTX; *p*aminobenzoic acid, *p*ABA; pan-assay interference compounds, PAINS; Potentiating Index, PI; Pteridine reductase 1: PTR1; Sinergy Index, SI; Short-chain dehydrogenase/reductase: SDR; Thymidylate Synthase, TS; *Trypanosoma brucei* pteridine reductase 1: *Tb*PTR1; *Trypanosoma brucei: T. brucei* or *Tb; Trypanosoma cruzi: T. cruzi;* Target Inhibitor Profile: TIP; Tetrahydrobiopterin: H₄B; van der Waals, vdW.

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Table 1. Chemical structures, enzymatic evaluation and anti-parasitic activity of etheric benzothiazoles 1a-g.

		R、		NH ₂			
Compound	D	IC ₅₀ i (%inhibitio	n μM ^a on at 50μM)	% cell growth inhibition [EC ₅₀ in μM]			
Compound	ĸ	<i>Tb</i> PTR1	LmPTR1	T. brucei ^b	Amastigote L. infantum ^b	Amastigote T. cruzi ^{a,c}	
1 a	-CH3	(29.9 %)	(15.2 %)	6±0	NI	32.9	
1b	-CF ₃	15.1	94.1	NI	NI	NI	
1c	*	13.4	NI	28±15	NI	NI	
1d		3.4	NI	42±11	NI	7.24	
1e	N N	0.67	86.5	27±0	NI	15.4	
1f		1.7	66.6	22±0	NI	8.79	
1g		3.3	31.9	52±0	NI	27.0	

a. Standard deviation is within $\pm 10\%$ of the value.

b. Compounds tested at 10μ M. NI: no inhibition.

 c. Compounds tested at 50 μ M . NI: no inhibition.

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Table 2. Chemical structures, enzymatic evaluation and anti-parasitic activity of sulfide benzothiazoles 2a-h.

R _S NH ₂								
Common d	D	IC ₅₀ i (%inhibitio	n μM ^a on at 50μM)	% cell growth inhibition [EC ₅₀ in μM]				
	K	<i>Tb</i> PTR1	LmPTR1	T. brucei ^b	Amastigote <i>L. infantum^b</i>	Amastigote T. cruzi ^{a,c}		
2a	-CH ₃	(16.7 %)	(26.9 %)	12±1	NI	22.5		
2b	-CF ₃	0.50	1.9	11±0	NI	37.4		
2c	*	1.9	81	14±17	NI	NI		
2d	CI CI	0.90	10	46±1	NI	21.4		
2e	N*	0.35	12.4	35±0	NI	27.3		
2f		0.92	47.6	17±0	NI	23.0		
2g		0.93	15.1	42±5	NI	16.3		
2h	H N N O	0.50	30	26±2	NI	19.2		

d. Standard deviation is within $\pm 10\%$ of the value.

e. Compounds tested at 10 μM . NI: no inhibition.

f. Compounds tested at 50 μM . NI: no inhibition.

Table 3. Chemical structures, enzymatic evaluation and anti-parasitic activity of sulfone benzothiazoles 3a,c-g.

R_{S} N_{NH_2}								
Compound	D	IC ₅₀ i (%inhibitio	n μM ^a on at 50μM)	% cell growth inhibition [EC ₅₀ in μM]				
Compound	K	<i>Tb</i> PTR1	LmPTR1	T. brucei ^b	Amastigote <i>L. infantum^b</i>	Amastigote T. cruzi ^{a,c}		
3a	-CH3	34.2	32.9	17±9	NI	17.4		
3c	*	57.6	(27.1 %)	5±17	NI	NI		
3d	CI CI	(16.7 %)	(26.9 %)	37±11	21±5	32.8		
3e		3.00	(55.5 %)	9±0	NI	NI		
3 f		15.1	65.9	14±1	12±17	NI		
3g		24.3	50	NI	NI	NI		

a. Standard deviation is within $\pm 10\%$ of the value.

 b. Compounds tested at 10 μM . NI: no inhibition.

c. Compounds tested at 50 μM . NI: no inhibition.

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Table 4. Chemical structures, enzymatic evaluation and anti-parasitic activity of 6-amide benzothiazoles 4c-t.

 $R^{-N} \rightarrow N = S^{-N} \rightarrow S^{-N} \rightarrow$

	D	D	R ₂ -	IC ₅₀ in μM ^a (%inhibition at 50μM)		% cell growth inhibition [EC ₅₀ in μM]		
Compound	K	\mathbf{R}_1		TbPTR1	LmPTR1	T. brucei ^b	Amastigote L. infantum ^b	Amastigote <i>T. cruzi^{a,c}</i>
4c	*	H-	H-	10.0	61.3	[7.0±0.5]	NI	51.3
4d	CI CI	H-	H-	5.4	15.3	5±8	5±5	NI
4e	N*	H-	H-	42.9	73.3	8±4	NI	NI
4f		H-	H-	16.5	36.0	16±2	NI	NI
4g		H-	H-	21.8	(1.97 %)	24±11	NI	NI
4h		H-	H-	6.85	18.9	53±0	NI	NI
4i		H-	H-	(30.8 %)	(67.1 %)	9±17	NI	19.8
4j	NO ₂	H-	H-	6.46	3.51	18±0	NI	19.0

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4k	NH ₂	H-	H-	13.9	64.7	8±0	NI	6.10
41		H-	H-	18.5	88.5	10±11	15±21	37.8
4m		*	H-	(36.4 %)	(23.3 %)	19±8	8±12	64.2
4n	s *	H-	H-	7.5	(18.3 %)	17±4	NI	29.8
40		H-	H-	23.3	NI	27±6	NI	1.64
4p		*	H-	12.0	NI	26±2	NI	62.1
4q		CH ₂ CH ₂ OH	H-	58.1	(24.1 %)	8±17	20±7	NI
4r	CI CI		H-	15.9	NI	[9.7±2.5]	9±8	47.9
4 s	*	H-	-OH	(47.8 %)	2.48	15±6	11±15	28.8
4t	CI CI	H-	-OH	(11.8 %)	(8.7 %)	[5.9±0.2]	24±15	42.0

a. Standard deviation is within $\pm 10\%$ of the value.

b. Compounds tested at 10μ M. NI: no inhibition.

 c. Compounds tested at 50 μM . NI: no inhibition.

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Table 5. Chemical structures, enzymatic evaluation and antiparasitic activity of 4-amide cathecolic benzothiazoles **5a-c.**



a. Standard deviation is within $\pm 10\%$ of the value.

b. Compounds tested at 10 μM . NI: no inhibition.

c. Compounds tested at 50 μM . NI: no inhibition.

SNAP-PK

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4c

TOC Graphic CHEMICAL AND BIOCHEMICAL PROPERTIES ANTIPARASITIC ACTIVITY EARLY-TOXICITY 2-AMINO-BENZOTHIAZOLES PROGRESSION OF COMPOUNDS BASED ON TARGET INHIBITOR PROFILE